

Development and Validation of Stability Indicating RP-HPLC Method for Estimation of Capmatinib in API Form

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ABSTRACT

A simple, rapid, precise, economical & accurate stability indicating RP-HPLC method for the estimation of Capmatinib in API form. High performance in liquid chromatographic. The separation was achieved by C18 (250 x 4.6mm, 5 μ) column and buffer (PH 6.5) : MeOH (60:40) as mobile phase, flow rate of 1.0 ml / minute. Detection is carried out at 231 nanometer. The drug was subjected

to forced degradation condition of acid degradation, base degradation, oxidation degradation, photodegradation and thermal degradation. It can be applied in routine analysis and pharmaceutical dosage forms. It can be applied in commercial pharmaceutical dosage forms. The ICH guidelines applicable for forced degradation studies are ICH Q1A, Q1B, Q2B.

KEYWORDS: Capmatinib, Stability, RP-HPLC, validation

I. INTRODUCTION

There are 2 main classifications of lung cancer: small cell lung cancer and non-small lung cancer (NSCLC). These 2 types are treated differently. NSCLC begins when healthy cells in the lung change and grow out of control, forming a mass called a tumour, a lesion, or a nodule. This can begin anywhere in the lung and the tumour can be cancerous or benign. When a cancerous lung tumour grows, it may shed cancer cells.

Types of non-small-cell lung cancer (NSCLC)

The different types of NSCLC are:

- Adenocarcinoma
- Squamous cell carcinoma
- Large cell carcinoma

Capmatinib is used to treat a **certain type of non-small cell lung cancer (NSCLC)** that has spread to other parts of the body. Capmatinib is in a class of medications called kinase inhibitors. It works by blocking the action of an abnormal protein that signals cancer cells to multiply. Capmatinib is in a class of medications

called **kinase inhibitors**.

It works by blocking the action of an abnormal protein that signals cancer cells to multiply. This helps slow or stop the spread of cancer cells. The mechanism of action of capmatinib is as a Mesenchymal Epithelial Transition Inhibitor, and Cytochrome P450 1A2 Inhibitor, and P-Glycoprotein Inhibitor, and Breast Cancer Resistance Protein Inhibitor, and Multidrug and Toxin Extrusion Transporter 1 Inhibitor, and Multidrug and Toxin Extrusion Transporter 2 K Inhibitor.

Sing Symptoms

Fatigue, Cough, Shortness of breath, Chest pain, if a tumor spreads to the lining of the lung or other parts of the body near the lungs, Loss of appetite, Coughing up phlegm or mucus, Coughing up blood, Unintentional weight loss, Horseness

Treatment

➤ There are 5 main ways to treat NSCLC:

- Surgery
- Radiation therapy
- Chemotherapy
- Targeted therapy and Immunotherapy

The forced degradation of any drug substance include:

1. Acid degradation
2. Base degradation
3. Oxidation degradation
4. Photo degradation
5. Thermal degradation

II. MATERIAL AND METHODS

1. Acid degradation:

Acid decomposition studies were performed by transferring one ml of stock solution in to 10 ml of volumetric flask. Two ml of 0.1 N HCl solution was added and mixed well and put for 2 hr at Room temperature. After time period two ml of 0.1m N NaOH Added to neutralize the solution and make up the volume with diluent to get 20 μ g/ml for Capmatinib.

2. Base degradation:

Basedecomposition studies were performed by transferring one ml of stock solution in to10 ml ofvolumetricflask.Twoml of0.1 N NaOHsolutionwasaddedandmixedwell andputfor 2 hr at Room temperature . After time period two ml of 0.1 N HCl Added to neutralize the solution andmake up thevolumewithdiluenttoget20 μ g/mlforCapmatinibOxidationdegradation:

Oxidationdecompositionstudies wereperformedbytransferringonemlofstocksolutioni nto10ml of volumetric flask. Two ml of 3% H₂O₂ solutions was added and mixed well and put for 2 hr at room temperature.Aftertimeperiodmakeupthevolumewith diluenttoget20 μ g/mlforCapmatinib.

3. Photodegradation:

Photo decomposition studies were performed by transferring one ml of stock solution in to 10 ml ofvolumetric flask. Volumetric flask

was kept in UV Chamber for 4 hrs. After time period make up thevolumewithdiluenttoget20 μ g/mlforCapmatinib.

4. Thermaldegradation:

20 mg of Capmatinib was taken in 100ml Volumetric flask and put in oven for 2 hrs at 80^oC temperature, Then after Volumetric flask was removed and cools at room temperature, volume was madeup with mobile phase, 1ml of this solution was transferred in 10ml volumetric and volume was madeupwithDiluentstoget20 μ g/mlforCapmatinib.

III. EXPERIMENTAL WORK

1. Acid Degradation: Capmatinib of Acid Degradation Blank is shown in **Figure 1**&Capmatinib of Acid Degradation Std. shown in **Figure 2**.

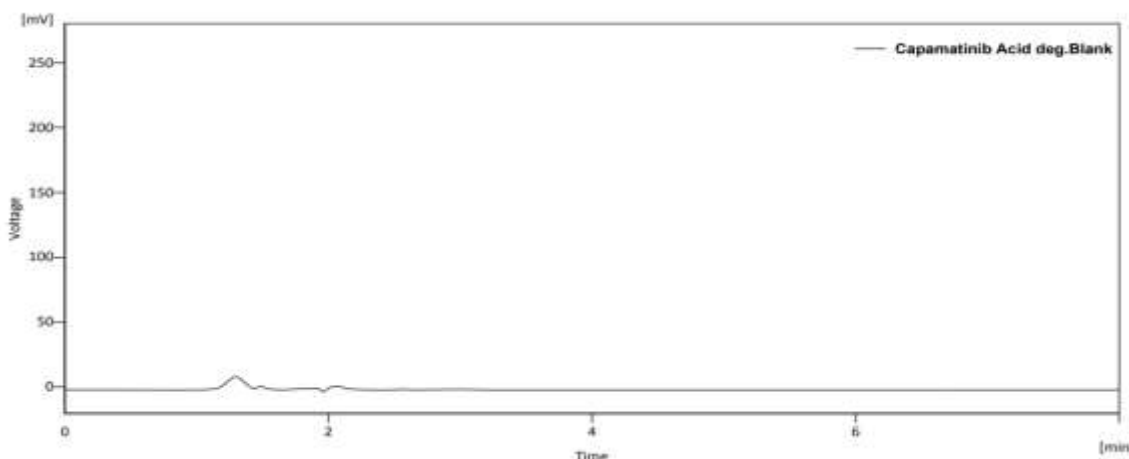


Figure 1: Capmatinib acid degradation of blank

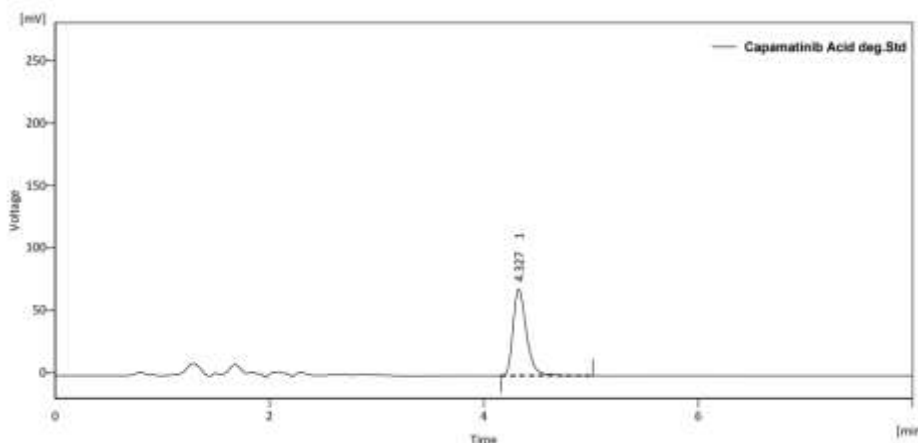


Figure 2: Capmatinib acid degradation of Standard

Column performance Table (From 50%-Capmatinib Acid deg.std)

No.	Reten. Time [Min.]	Asymmetry[-]	Efficiency[th.pl]	Resolution [-]
1.	4.327	1.533	6173	-

2. **Base Degradation:**Capmatinib of Base Degradation Blank is shown in **Figure 3**&Capmatinib of Base Degradation Std. shown in **Figure 4**

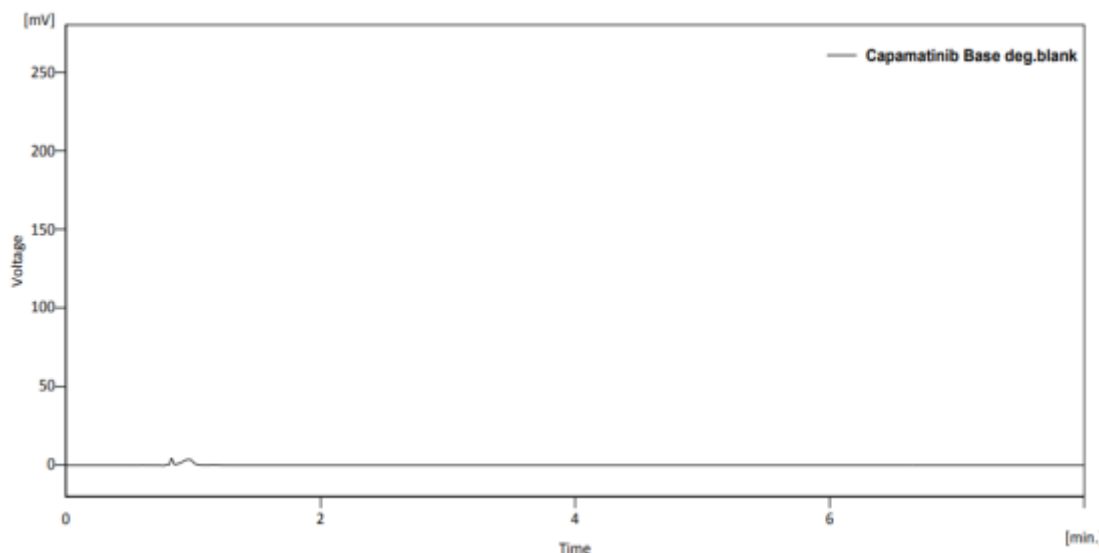


Figure 1:Capmatinib Of Base Deg. Blank

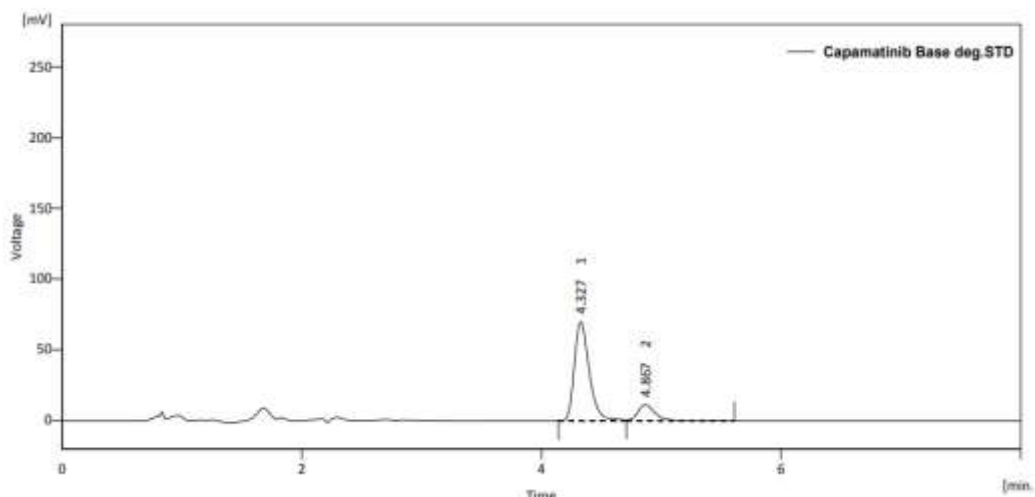


Figure 2: Capmatinib base degradation Standard

Column Performance Table (From 50%-Capmatinib Base deg.std)

	Retention Time (min)	Asymmetry (-)	Efficiency (th.pl)	Resolution (-)
1	4.327	1.533	5834	-
2	4.867	1.486	5832	2.243

3. **Oxidation degradation:**Capmatinib of Oxidation Degradation Blank is shown in **Figure 5**&Capmatinib of Oxidation Degradation Std. shown in **Figure 6**

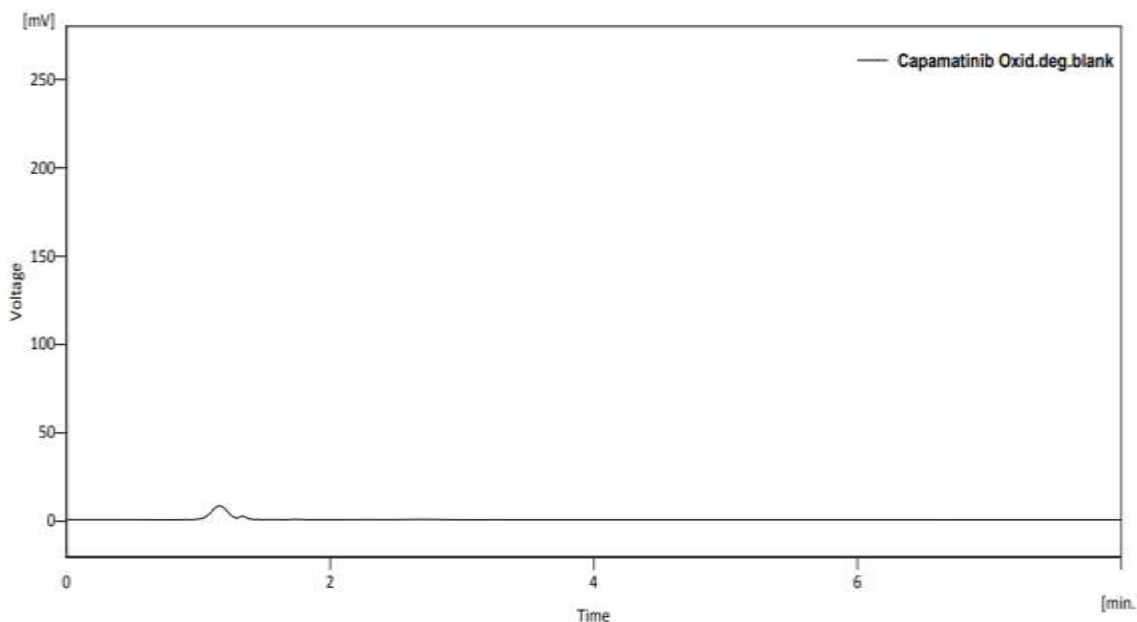


Figure 5:Oxidation Degradation blank

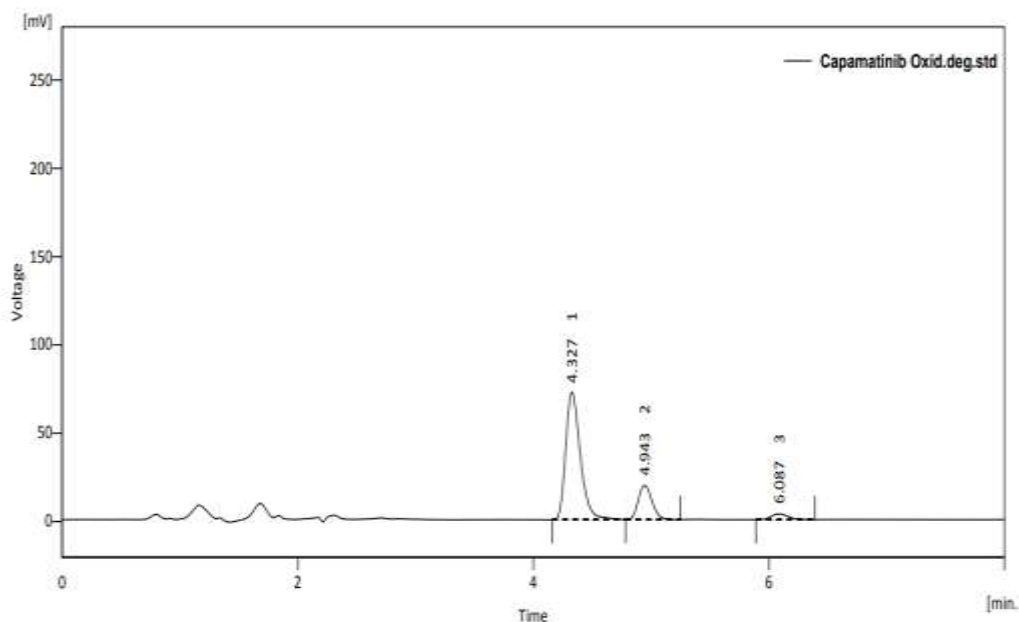


Figure 6 :Chromatogram of Oxidation Degradation Standard

Column Performance Table (From 50% -CapmatinibOxi.deg.std)

No.	Retension Time (min)	Asymmetry (-)	Efficiency (th.pl)	Resolution (-)
1	4.327	1.533	5834	-
2	4.943	1.333	8011	2.765
3	6.087	1.297	8362	4.694

4. **Photo degradation:**Capmatinib of Photo Degradation Blank is shown in **Figure 7**&Capmatinib of Photo Degradation Std. shown in **Figure 8**

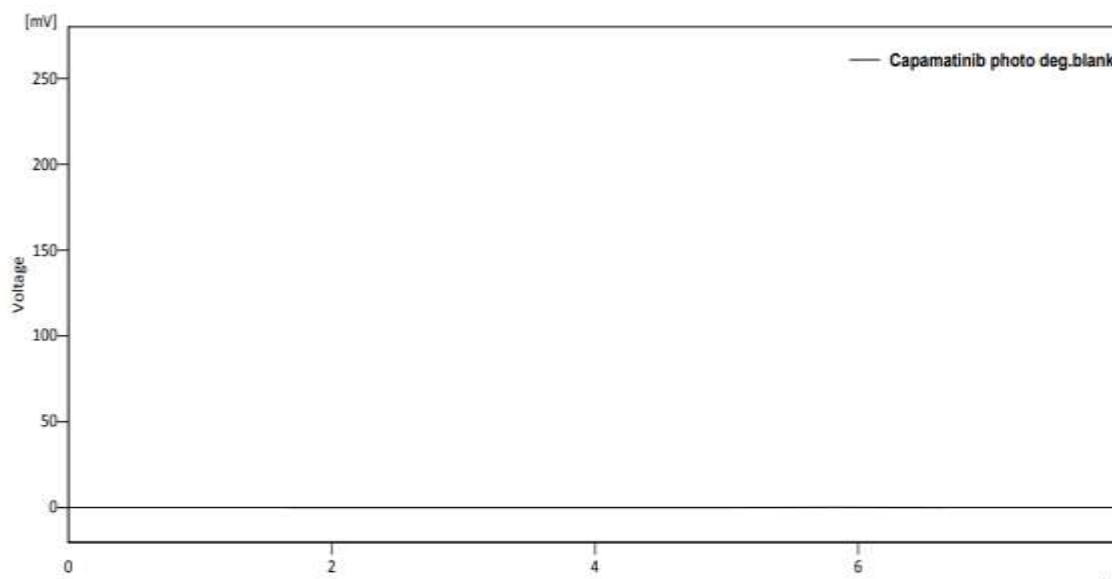


Figure 7: Chromatogram of Photo Degradation Blank

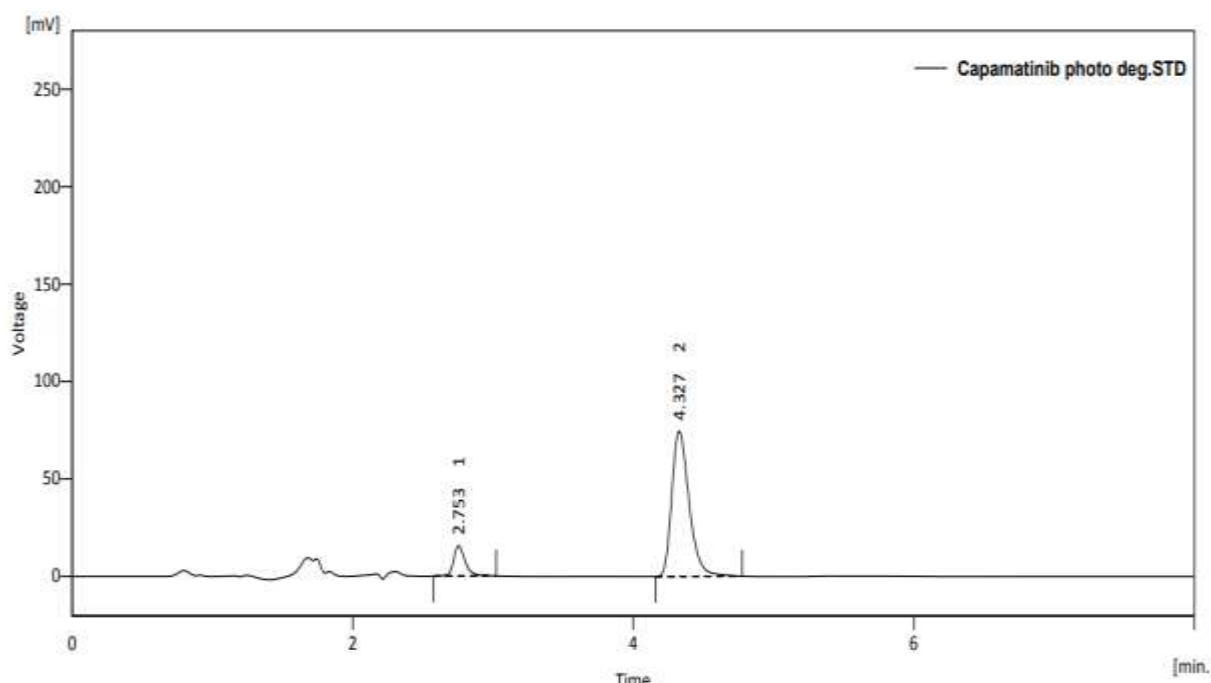


Figure 8: Chromatogram of Photo Degradation Standard

Column Performance Table (From 50% -Capmatinib Photo deg.std)

	Retention Time (min)	Asymmetry (-)	Efficiency (th.pl)	Resolution (-)
1	2.753	1.550	5591	-
2	4.327	1.533	5834	8.416

5. **Thermal Degradation:** Capmatinib of Thermal Degradation Blank is shown in **Figure 9** & Capmatinib of Thermal Degradation Std. shown in **Figure 10**.

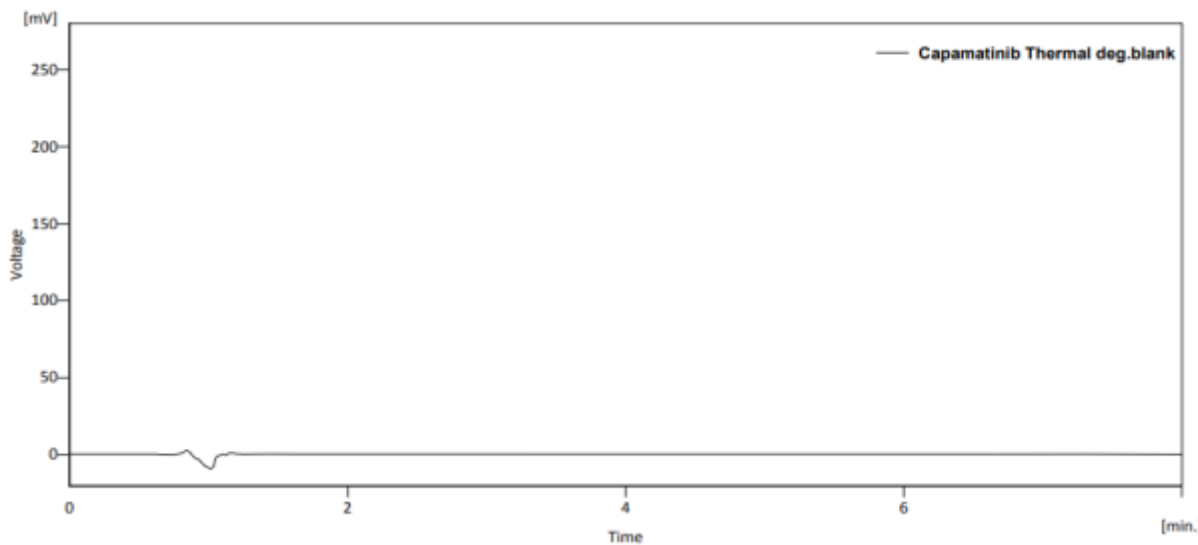


Figure 9: Chromatogram of Thermal Degradation Blank

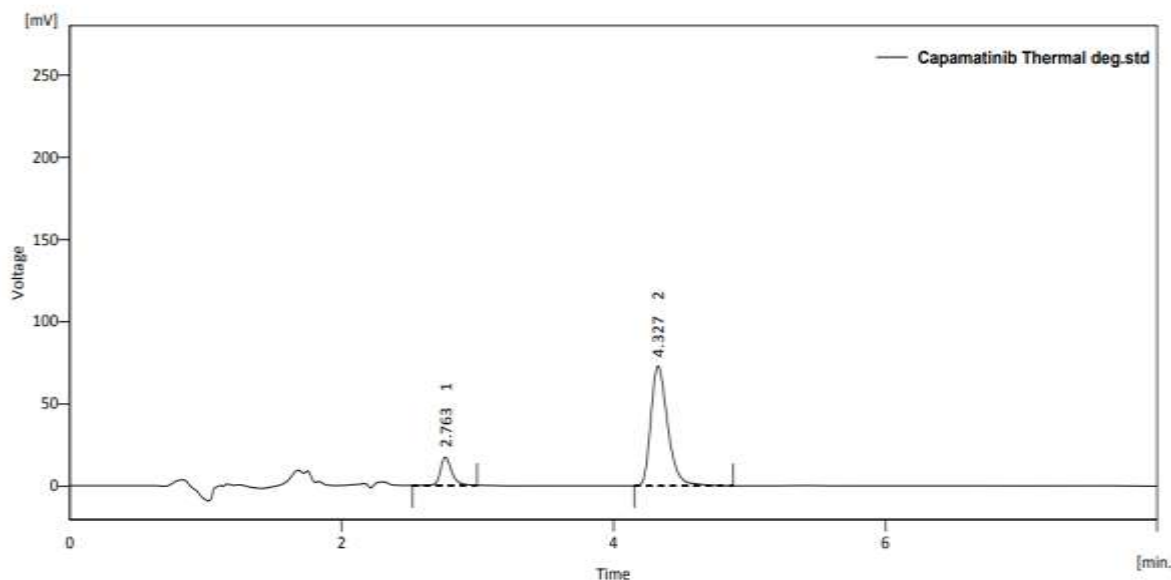


Figure 10: Chromatogram of Thermal Degradation Standard

Column Performance Table (From 50%-Capmatinib Thermal deg.std)

No.	Retention Time (min)	Asymmetry (-)	Efficiency (th.pl)	Resolution (-)
1	2.763	1.476	5632	-
2	4.327	1.533	5834	8.363

IV. RESULT:

1. Acid Degradation:

Table 1: (Uncal – Capmatinib Acid degradation Std)

No.	Reten.Time [Min.]	Area[mV.s]	Area [%]	Height [mV]
1.	4.327	596.316	100.0	69.155
	Total	596.316	100.0	69.155

2. Base Degradation:

Table 2 : (Uncal-Capmatinib Base degradation Std)

No.	Reten.Time [Min.]	Area[mV.s]	Area[%]	Height[mV]
1.	4.327	601.391	84.2	69.843
2.	4.327	112.482	15.8	11.279
	Total	713.873	100.0	81.121

3. Oxidation Degradation:

Table 3: (Uncal-Capmatinib Oxi.deg.Std)

No.	Reten.Time [Min.]	Area[mV.s]	Area[%]	Height[mV]
1.	4.327	621.441	76.7	72.106
2.	4.943	156.883	19.4	19.386
3.	6.087	32.046	4.0	3.271
	Total	810.370	100.0	94.763

4.

Photo Degradation:

Table 4: (Uncal-Capmatinib Photo deg. Std)

No.	Reten.Time [Min.]	Area[mV.s]	Area[%]	Height[mV]
1.	2.753	88.640	12.1	15.539
2.	4.327	640.919	87.9	74.584
	Total	729.558	100.0	90.123

5. Thermal Degradation:

Table 5: (Uncal-Capmatinib Thermal deg.Std)

No.	Reten.Time [Min.]	Area[mV.s]	Area[%]	Height[mV]
1.	2.763	98.836	13.6	17.190
2.	4.327	628.178	86.4	72.867
	Total	727.014	100.0	90.057

CONCLUSION:

Forced degradation studies give knowledge on possible degradation pathways and degradation products of the API and help explain the structure of the degradants. Degradation products cause from forced degradation studies are helpful possible degradation products which may or may not be applicable under storage conditions but they help in the developing stability indicating method. It helps in drug development process and the stability of the molecule. This information will further help improve the formulation manufacturing process and access to storage

conditions. The aim of strategy used for forced degradation is to create the required amount of degradation i.e., 5–20%. A properly planned and performed forced degradation study is used to generate proper sample for development of stability indicating method.

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