

## Microsphere: A Review Article

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### ABSTRACT

A controlled drug delivery system can improve a medicine's therapeutic efficacy and overcome the problems associated with traditional drug therapy.

As an efficient therapeutic substitute for traditional or immediate-release single-unit dosage forms, microspheres are one of the innovative drug delivery systems. Spherical, free-flowing powders with particles smaller than 200  $\mu\text{m}$  are called microspheres.

Made up of artificially produced proteins and polymers that naturally decompose. Microspheres increase stability, decrease dose frequency, improve absorption, lessen side effects, and deliver the medication to a precise location at a predefined pace. Microspheres come in a variety of forms, including biodegradable, polymeric, radioactive, floating, and bioadhesive. In the future, microspheres will locate the target in innovative medication delivery systems, particularly in genetic materials, diagnostics, targeted and efficient medicine distribution.

**Keywords:** Microspheres, Microsphere Types, Microsphere Characterization, Preparation Techniques, Use of microspheres.

### I. INTRODUCTION;

Microspheres are tiny spherical particles with dimensions ranging from micrometers to (usually between 1  $\mu\text{m}$  and 1000  $\mu\text{m}$ ). Microparticles are another name for microspheres. Numerous synthetic and natural materials can be used to create microspheres. There are three types of commercially accessible microspheres: glass, polymer, and ceramic. Because of their vastly varied densities, solid and hollow microspheres have various uses. Usually, hollow microspheres are added to materials to reduce their density. These solid microspheres can be used for a variety of purposes, depending on their size and composition. Microspheres made of polyethylene and polystyrene are the two most prevalent kinds of polymers. tiny spheres. Usually employed in biomedical applications, polystyrene microspheres have the capacity to make processes like immunoprecipitation and cell sorting easier. Because proteins and ligands may

easily and permanently adsorb onto polystyrene, polystyrene microspheres can be used in biological laboratory studies and medical research. Both temporary and permanent fillers are frequently made of polyethylene microspheres. Because polyethylene microspheres melt at a lower temperature, they can form porous structures in ceramics and other materials. Polyethylene microspheres are very popular for fluid flow studies and flow visualization, microscopy methods, health sciences, and other applications because to their high sphericity and availability of colored and fluorescent microspheres. process debugging as well as a variety of research uses. Additionally, electronic paper uses charged polyethylene microspheres. electronic screens.

Glass microspheres have few uses in medical technology and are mostly employed as fillers to reduce weight, retro-reflectors to improve highway safety, and additives for cosmetics and adhesives. The main application for ceramic microspheres is as grinding media.

The quality, sphericity, uniformity, and size distribution of microspheres varies greatly. It is necessary to select the proper microsphere for every distinct application. Numerous options to regulate various aspects of medication administration are presented by the multitude of methods available for creating microspheres. This method makes it easier to distribute small amounts of powerful medications precisely, lower drug concentrations at locations other than the target site, and protect labile compounds. both before and after the administration, as well as before showing up to the activity site. The way the medications behave in vivo can be controlled through the drug's binding to a carrier particle. The behavior of the carrier has a significant impact on the drug's clearance kinetics, tissue distribution, metabolism, and cellular contact. Utilizing these modifications in pharmacodynamic behavior could result in a more effective treatment outcome. However, a thorough comprehension of the carrier interaction is necessary for an intelligent treatment approach utilizing drug carrier technology. Drugs can be controlled in vivo by attaching them to a carrier particle. The behavior of the carrier has a

significant impact on the drug's clearance kinetics, tissue distribution, metabolism, and cellular contact. Taking advantage of these A better therapeutic impact could result from modifications in pharmacodynamic behavior. Any drug delivery system's objective is to deliver a therapeutic dose of medication to the appropriate location in the body in order to quickly reach and then sustain the required drug concentration. Oral consumption has long been the most practical and widely used method of medication administration. Medications with a short half-life and easy absorption from the GIT are rapidly removed from the bloodstream. To stay clear of these issues oral regulated medication administration mechanisms that maintain a steady drug concentration in the serum and deliver the medication gradually into the GIT for longer duration. Lower bioavailability, however, will result from the drug's partial release and the dosage forms' shorter residence time in the upper gastrointestinal tract, a major site for drug absorption. As the pharmaceutical business has expanded, so too have efforts to increase the

bioavailability of oral drugs. New approaches are needed to create oral active treatments because of the rise in medication quantity and chemical variety. As a result, gastro retentive dosage forms have been created, which increase the medications' bioavailability and lengthen their stay in the stomach<sup>2</sup>. A carefully thought-out, regulated medication distribution system can get beyond some alleviate the drawbacks of traditional treatment and improve a medication's therapeutic effectiveness. To acquire For best therapeutic efficiency, the agent must be delivered to the target tissue in the ideal quantity within the ideal time frame, resulting in low toxicity and negligible side effects. Delivering a medicinal ingredient to the target site in a prolonged controlled release form can be accomplished in a number of ways. Using microspheres as medication carriers is one such strategy. The distinctive free-flowing powders known as microspheres are made of proteins or synthetic polymers that are biodegradable and preferably have a particle size of less than 200  $\mu\text{m}$ <sup>3</sup>.

## CHARACTERISTICS

Table 1: Microsphere property

S. No.	Property	Consideration
1	Size	Diameter Uniformity/distribution
2	Composition	Density Refractive index Hydrophobicity/hydrophilicity Nonspecific binding Autofluorescence
3	Surface chemistry	Reactive groups Level of functionalization Charge
4	Special properties	Visible dye/fluorophore Super-paramagnetic

1. Microsphere size may be secondary to other features or essential to an assay's correct operation. When using conventional diagnostic techniques, the format of the test or assay typically determines particle size, like using tiny spheres (~0.1–0.4 $\mu\text{m}$ ) to guarantee adequate wicking in lateral flow experiments, or using bigger, cell-sized spheres (about 4–10  $\mu\text{m}$ ) for flow cytometric tests based on beads.
2. Polystyrene (PS), poly(methyl methacrylate) (PMMA), and silica are common microsphere compositions. These materials have various optical and physical characteristics, which could be advantageous or disadvantageous for particular uses. Because polymer beads are often hydrophobic, they have a high capacity to bind proteins. To ensure handling ease, they frequently call for the addition of a surfactant (such as 0.01-0.1% Tween® 20 or SDS) to the storage buffer. Beads with surface reactive

- groups can be created during synthesis by co-polymerizing functional monomers with methyl methacrylate or styrene. In addition to helping to stabilize the suspension, functional groups can be employed in covalent binding reactions. Microspheres of silica are fundamentally negatively charged and hydrophilic. As a result, surfactants or other stabilizers are rarely needed for aqueous silica suspensions. Common covalent coating procedures can employ carboxyl- and amine-functionalized silica spheres, while a range of silanes can be used to modify plain silica microspheres in order to create functional groups or change surface characteristics.
3. For usage in diagnostic or separation applications, microspheres can be coated with capture molecules including peptides, oligonucleotides, antibodies, etc. Usually, microsphere coatings are tuned to minimize

nonspecific interactions while achieving the required specific activity. Additionally, the necessary stability, the budget and schedule for development, and the particular biomolecule to be covered. These elements will help choose the best coating approach for both immediate and long-term goals. Three fundamental coating techniques are supported by standard microsphere products: affinity binding, covalent coupling, and adsorption.

4. Additional characteristics, like fluorescence or a visible color, or iron oxide inclusions for magnetic separations, are required for numerous applications in the life sciences. Many conventional goods are available, and polymer spheres (and polymer-based magnetic spheres) are frequently internally dyed via organic solvent swelling. Concentrations of dyes can be adjusted to produce beads with different intensities to meet special needs, such as QuantumPlex™ for multiplexed flow cytometric assays, or our Dragon Green or Flash Red Intensity Standards, which support imaging applications and associated instrument QC. Many surface- or internally labeled fluorescent beads are also available as specialized flow cytometry standards.<sup>9</sup>

#### ADVANTAGES

1. Microspheres have a long-lasting and consistent therapeutic impact.
2. Improves patient compliance by lowering the frequency of dose.
3. Because of their smaller size and spherical shape, they may be injected into the body.
4. Improved medicine use will increase bioavailability and lessen the frequency or severity of side effects.
5. The shape of microspheres permits controlled variation in medication release and breakdown.<sup>7</sup>

#### LIMITATIONS OF MICROSPHERES

- Drug release from various formulations varies through the GIT,
- The rate of release varies from dosage to dose.
- A non-uniform drug release could result in toxicity.
- The tablet should not be crushed or chewed.
- The modified release from the formulations.

#### TYPES OF MICROSPHERES:

1. Bioadhesive microspheres
2. Magnetic microspheres
3. Radioactive microspheres

4. Floating microspheres
5. Polymeric microspheres
6. Biodegradable polymeric microspheres
7. Synthetic polymeric microspheres

#### Bioadhesive microspheres

Adhesion is the process by which a medication adheres to a membrane by utilizing the ability of water-soluble polymers to stick. Bioadherence can be defined as adhesion to the mucosal membrane of the drug delivery device, including the buccal, ocular, rectal, nasal, etc. These types of microspheres show an extended stay at the application site, elicit close contact with the absorption site, and produce improved therapeutic results. Mucoadhesive microspheres promote close contact and offer prolonged contact time at the application or absorption site. The underlying surface where absorption is anticipated to take place, increasing the drug's therapeutic efficacy.

#### Magnetic microspheres

Microspheres are typically free-moving, tiny, spherical particles with a size range of 1–1000 µm that are made of proteins or synthetic polymers and are biodegradable. They are regarded as one of the most important methods for safely administering therapeutic substances and regulated way for delivery to the intended location.

The various kinds of

##### a) Therapeutic magnetic microspheres

renowned for treating liver tumors with chemotherapy. Medicinal items like proteins and peptides can also be targeted by this technology.

##### b) Diagnostic Microspheres

By producing supramagnetic iron oxide particle nanometers, it can be utilized to visualize liver metastases and to differentiate intestinal loops from other abdominal structures.

#### Radioactive microspheres

Radio embolization treatment microspheres are 10–30 nm larger than capillary diameters and are inserted into the first capillary bed as they pass. In each of these circumstances, they are inserted into the arteries that lead to the tumor of interest and cause increased radioactivity. microspheres to target regions without causing damage to the normal tissues that surround them. This differs from the medication delivery method in that radioactivity is not released from microspheres but rather functions from a distance characteristic

of a radioisotope. There are three types of radioactive microspheres:  $\alpha$ ,  $\beta$ , and  $\pi$  emitters.

### Floating Microspheres

Floating forms remain buoyant in the stomach without affecting the rate of gastric emptying because their bulk density is lower than that of the gastric fluid. As the body floats on stomach content and reduces gastric residency and plasma, the medication is progressively released at the desired rate. variation in concentration. Additionally, this lessens the possibility of striking and dose dumping. One way it lowers the frequency of dose is by producing a long-lasting therapeutic impact.

### Polymeric microspheres

Biodegradable and synthetic polymeric microspheres are the two categories into which the various kinds of this material can be divided.

### Biodegradable polymeric microspheres

Because they are inherently biodegradable, biocompatible, and even bioadhesive, natural polymers like starch are employed. Because of their strong swelling capabilities with aqueous media, biodegradable polymers prolong their time in contact with the mucosal membrane. The Polymer concentration and release pattern sustainably control the rate and extent of medication release. The main drawback is that controlling medication release and ensuring the dependability of biodegradable microspheres in clinical settings are challenging. Nonetheless, they provide a broad range of microsphere-centered therapy applications.

### Synthetic polymeric microspheres

Although synthetic polymeric microspheres have been shown to be safe and biocompatible and are frequently employed in therapeutic applications as bulking agents, fillers, embolic particles, drug delivery vehicles, etc., their primary disadvantage is that they seem to move away from the injection site, which could cause injury, embolism, and additional organ damage.

### Materials used in the microsphere formulation

In the formulation of microsphere mainly used a polymers, they are classified as follows.

- Synthetic Polymers
- Natural polymers

#### A.Synthetic polymers are divided into two types

- a) Non-biodegradable polymers  
Example- Poly methyl methacrylate (PMMA), Acrolein  
Glycidyl methacrylate, Epoxy polymers
- b) Biodegradable polymers-  
Example- Lactides, Glycolides and their co polymers,  
Poly alkyl cyano acrylates, Poly anhydrides

#### B. Natural polymers-

They are obtained from different sources like proteins, carbohydrates and chemically modified carbohydrates. They are also used a protein like Albumin, Gelatin, and Collagen, Carbohydrates like Agarose, Carrageenan, Chitosan, Starch and also Chemically changed carbohydrates used like Poly dextran, Poly starch.

**Single emulsion technique:**

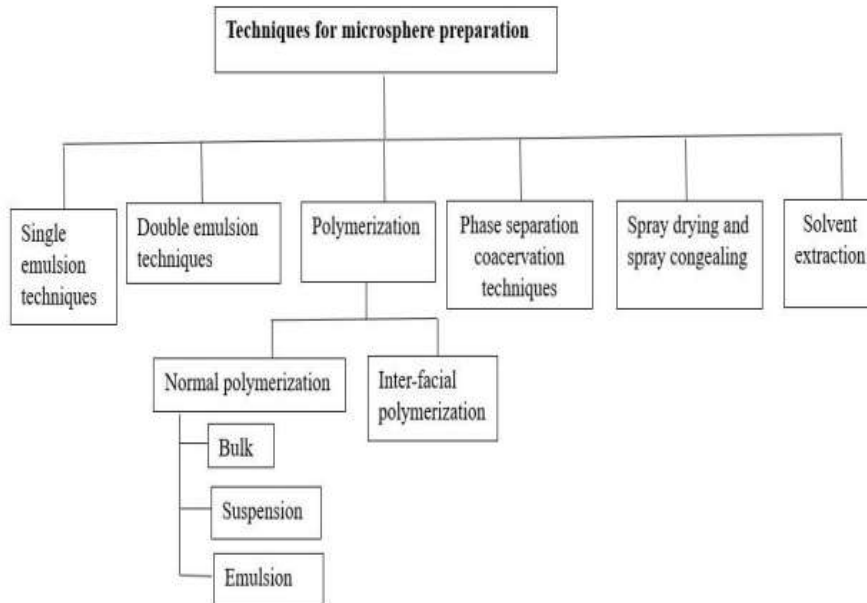


Figure 1: Technique for Microspheres preparation<sup>10</sup>

**Single emulsion technique:**

When made utilizing a single emulsion method, natural polymers like proteins and ex-carbohydrates can act as microparticulate carriers. Oils and other non-aqueous media dissolve or disperse natural polymers. Cross connecting is then

done using one of the two techniques listed below:18.

**3.1.1 Cross linking by heat:** This method involves adding the dispersion to heated oil for cross linking. However, heat denaturation is inappropriate for medications that are thermolabile.



Figure 2: Single emulsion technique<sup>13</sup>

**3.2 Double emulsion technique:** When adding water-soluble medications, peptides, proteins, and vaccines to microspheres, the double emulsion solvent evaporation/extraction technique works well. This entails spreading adding polyvinyl

alcohol to a homogenized protein solution in a lipophilic organic continuous phase to create a double emulsion. After that, the emulsion is eliminated by solvent extraction or evaporation, producing solid microspheres. The incorporation of

hydrophilic medications, vaccines, proteins/peptides, and conventional compounds has

been accomplished with success using this technique.

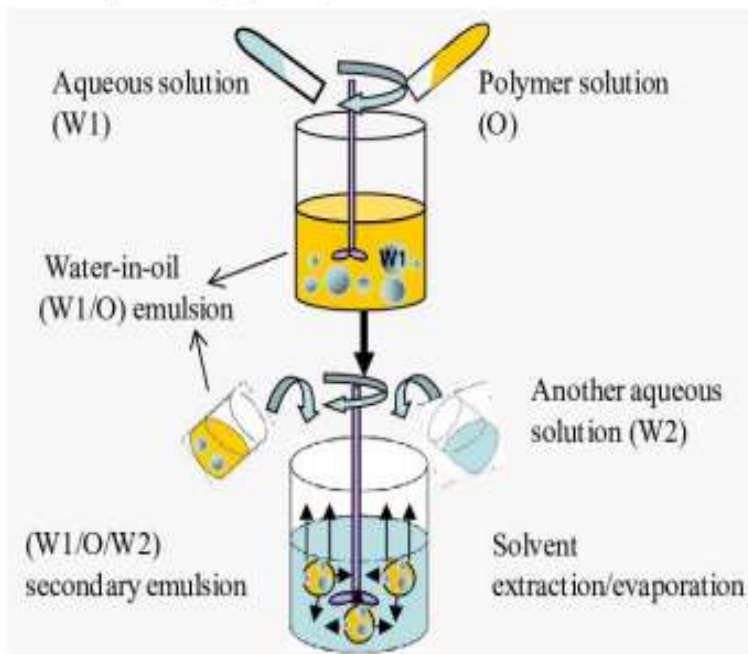


Figure 3: Double emulsion technique<sup>14</sup>

**3.3 Polymerization technique:** Several polymerization techniques are used to create microspheres, such as:

- Regular polymerization
- Interfacial polymerization

**3.3.1 Normal polymerization:**

**Bulk polymerization:** In order to initiate and finish the polymerization process, a monomer or a monomer and initiator combination is frequently heated. The catalyst or initiator is added to the reaction mixture to aid in or accelerate the process. The resultant polymer can be shaped or broken up into microspheres. Two methods for drug loading include adsorption-based loading and drug addition during the polymerization process.

**The suspension polymerization:** It is accomplished by heating the monomer or mixture of monomers that contain active components (drugs) as droplets dispersed in a continuous

aqueous phase. The droplets could possibly contain an initiator and additional chemicals.

However, emulsion polymerization differs from suspension polymerization in that the initiator is present in the aqueous phase and subsequently diffuses to the surface of the micelle or emulsion globule<sup>17</sup>.

**3.3.2 Interfacial polymerization:** The interfacial polymerization process involves two reactive monomers, one dissolved in the continuous phase and the other distributed there. The second monomer is emulsified during the continuous phase, often aqueous. The monomers diffuse quickly and polymerize quickly at the interface. The polymer's solubleness in the emulsion droplet can affect the carrier form. Temperature, vehicle composition, monomer concentration, and reactivity can affect polymerization. Particle size can be regulated by adjusting the size of dispersed phase droplets or globules. Controlling the polymerization process requires maintaining monomer concentration.

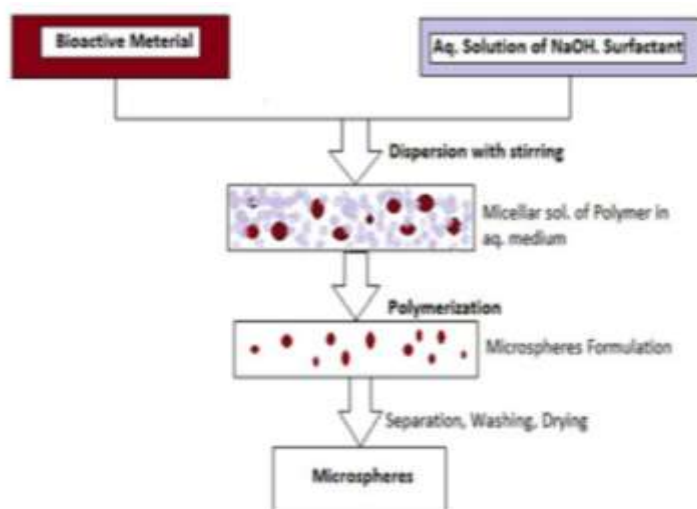


Figure 4: Polymerization technique<sup>14</sup>

### 3.4 Phase separation coacervation technique:

designed especially to create the reservoir kind of the system, which encapsulates drugs that are hydrophobic, like steroids, and soluble in water, such proteins and peptides. In a matrix-type device, the drug or protein dissolves in the polymer phase. The technique influences the formation of the coacervates, a phase rich in polymers, by decreasing the solubility of the polymer in the organic phase. Adding a third component to the

system can exacerbate the situation by causing the production of two phases, one of which is the supernatant depleted of polymer. This technique entails dissolving the polymer in the proper solvent first, followed by the drug's dispersion in an aqueous solution if it is hydrophilic or hydrophobic, by dissolving it within the actual polymer solution. Phase separation is then accomplished by modifying the solution's conditions<sup>21,18</sup>.

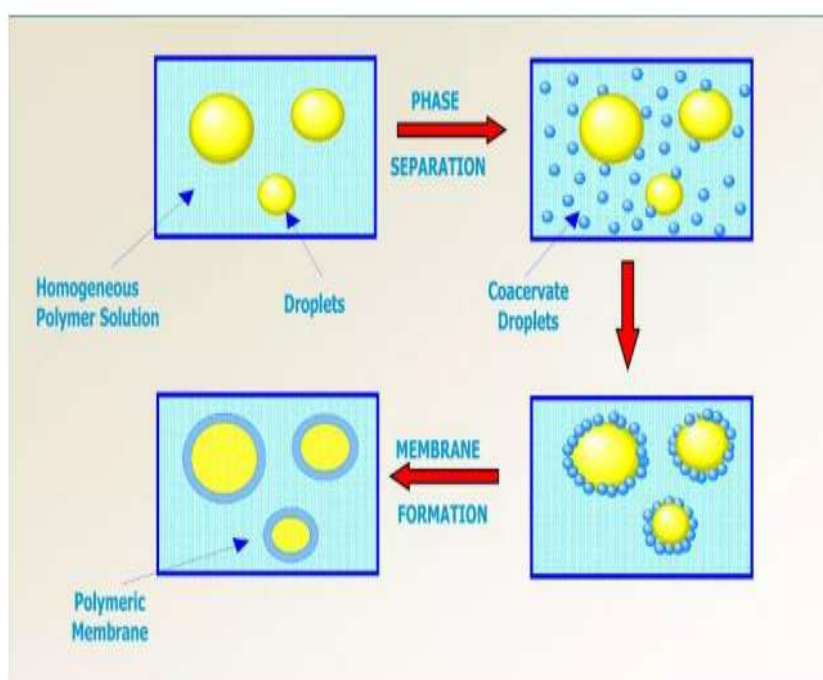


Figure 5: Phase separation coacervation technique<sup>17</sup>

**3.5 Spray drying and spray congealing:** When the solution cools or the solvent is eliminated, two processes take place: spray drying and spray congealing. Evaporation is the basic process of spray drying, whereas a phase inversion from a liquid to a solid is the mechanism of spray congealing. The only difference between the two processes is energy flow. Spray drying is the most popular industrial technique for drying and shaping particles. Spray drying is therefore the ideal technique when the finished product must satisfy precise specifications for bulk density, particle shape, residual moisture content, and particle size distribution.

The idea There are three stages to spray drying:

- Atomization: the transformation of a liquid stream into tiny droplets.

- Mixing: this process includes directing a hot gas stream through spray droplets, causing liquids to evaporate and leaving behind dry particles.
- Dry: The powder is collected after being dried and removed from the gas stream.

Double emulsion solvent evaporation/extraction is the best technique for adding water-soluble medications, proteins, peptides, and vaccinations to microspheres. A protein solution is dispersed inside a lipophilic organic continuous phase, creating a double emulsion by homogenizing it and then adding polyvinyl alcohol. Solid microspheres are the end result of solvent extraction or evaporation, which removes the emulsion. Conventional chemicals, proteins/peptides, hydrophilic medications, and vaccinations have all been successfully included using this technique.

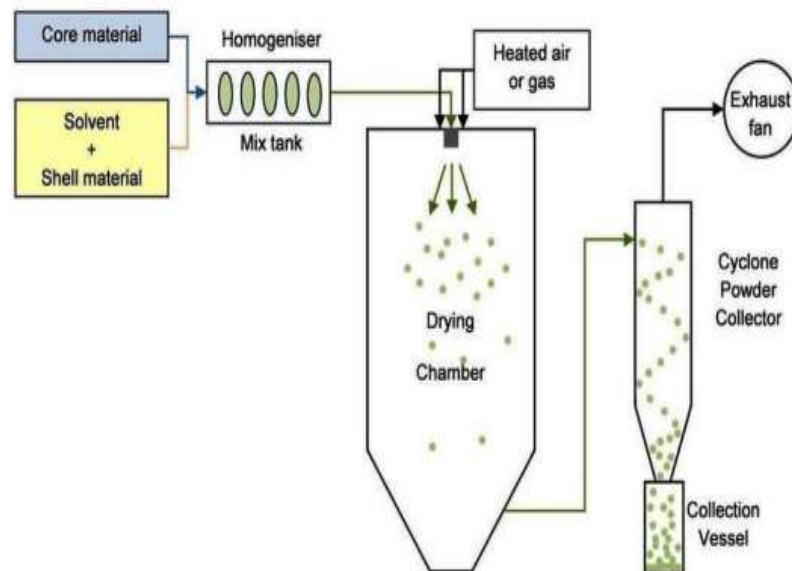


Figure 6: Spray drying and spray congealing<sup>20</sup>

**3.6 Solvent extraction:** for the emulsion to form in both the aqueous (o/w) and non-aqueous (w/o) phases between the polymer solution and an immiscible continuous phase. Bogataj et al. employed the evaporation process in their 2000 study to produce microspheres using acetone and liquid paraffin as solvents. Following the dispersion of the medicine solution (in acetone) in the chitosan solution, the mixture was stirred and emulsified in liquid paraffin. The suspension of

microspheres was filtered, cleaned, and dried. Magnesium stearate was also employed as an agent to avoid agglomeration. The results showed that the average particle size decreased as more magnesium stearate was used to create the microspheres. Compared mucoadhesive microspheres of hyaluronic acid, chitosan glutamate, and a combination with hyaluronic acid and gelatin microcapsules created via complex coacervation.



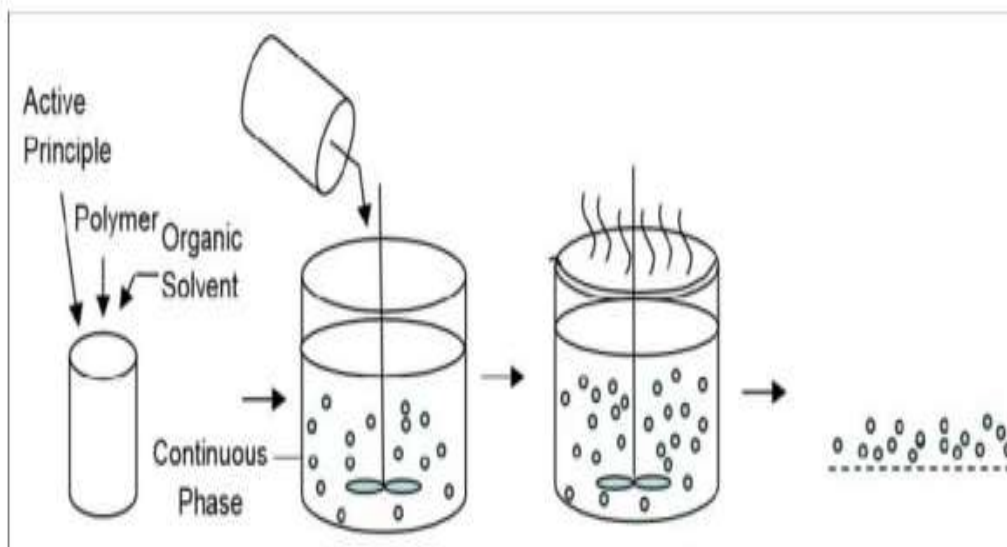


Figure 7: Solvent Extraction Technique<sup>21</sup>

#### 4. Evaluation of Microspheres:

**4.1 Particle size and shape:** Microparticles are frequently studied using scanning electron microscopy (SEM) and conventional light microscopy (LM), which show their exterior shape and structure. Higher resolution is provided by SEM, but flexibility over coating parameters is possible with LM. Multiple-walled microspheres are characterized by confocal fluorescence microscopy, whereas laser light scattering and multisize Another option is to use a Coulter counter<sup>26</sup>.

**4.2 Electron spectroscopy for chemical analysis:** Confocal fluorescence microscopy assesses the structural characteristics of multiwalled microspheres. Multisize microspheres can be analyzed for size, shape, and morphology in addition to instrumental methods. Coulter Counter and Scattering of Laser Light <sup>27, 38</sup>.

#### 4.3 Attenuated total reflectance Fourier Transform Infrared Spectroscopy:

FT-IR is used to evaluate the degradation of the polymeric matrix in the carrier system. The microspheres' surface is mostly measured using alternate total reflectance, or ATR. The sample's surface material's infrared spectra were mostly acquired by several

IR beam reflections that went via the ATR cell.

Information on surface composition is obtained using ATR-FTIR analysis. About the microspheres according to conditions and manufacturing methods.

**4.4 Density determination:** The density of microspheres can be measured using a multivolume pycnometer. A properly weighed sample is put in a cup and placed into the multivolume pyrometer. Helium at constant pressure is poured into the chamber and allowed to expand. This expansion causes the pressure inside the chamber to drop. Two successive pressure drop readings are captured, each starting at a different pressure. The volume and, thus, the density of microsphere carriers<sup>28</sup> are determined using two pressure measures.

**4.5 Isoelectric point;** To determine the isoelectric point of microspheres, their electrophoretic mobility is assessed using a device known as a micro electrophoresis. The mean velocity is calculated at different pH values between 3 and 10 by timing particle movement over a distance of 1 mm. The electrical mobility of the particle can be ascertained using this information. Microspheres' surface charge, ionisable behavior, and ion absorption all affect their electrophoretic mobility<sup>27,37</sup>.

**4.6 Drug entrapment efficiency:** A measured quantity of microspheres is removed and broken apart. then, with the aid of a stirrer, dissolved in buffer solution and filtered. Using a calibration curve, the filtrate is tested at a certain wavelength using a UV spectrophotometer <sup>39</sup>.

Drug Entrapment efficiency =  $\frac{\text{Actual weight of microspheres}}{\text{Total weight of microspheres}}$

Theoretical wt.of drug  
and polymer $\times$  100

**4.7 Percentage yield:** It is computed by dividing the total weight of the medicine and polymer needed to make each batch by the weight of microspheres that were obtained from it, then multiplying the result by 100

**4.8 Swelling index:** It is ascertained by measuring the degree of microsphere swelling in a certain solvent. To achieve equilibrium swelling, five milligrams of dried microspheres are mixed with five millilitres of buffer solution and kept overnight in a measuring cylinder. It is computed using the provided formula.

**4.9 In vitro techniques:** This method assesses a drug's properties of release and membrane permeability. The in vitro method is a quality control technique used in pharmaceutical manufacture, product development, and other fields. Sensible and reproducible release data produced from chemically, physically, and hydrodynamically defined settings are crucial<sup>26,27</sup>.

#### 4.10 Interface diffusion method:

compartment C, which represents bodily fluids. One octanol is also present in compartment D, which symbolizes protein binding. The 1-octanol and aqueous phases are saturated with one another before to use. The samples are taken out and placed back into compartment A using a syringe<sup>18,36</sup>.

**4.11 In vivo method:** Techniques that provide direct measurements of drug absorption or accumulation at the surface, as well as biological reactions locally or systemically, are used to assess the permeability of intact mucosa. Animal models and buccal absorption tests<sup>20</sup> are commonly used in in vivo research.

## 5. Microsphere application:

### 5.1 Vaccine delivery using microspheres:

Immunity to the microorganism or any of its toxic metabolites is necessary for a vaccine. Affordable, safe, simple to use, and effective are the qualities that the ideal vaccination should have. Reducing adverse reactions and ensuring safety are two intricate problems. Both the safety factor and the degree of antibody response are directly impacted by the injection method. The use of biodegradable delivery systems for parenteral vaccinations is one possible way to alleviate the drawbacks of conventional vaccines. Parenteral

(subcutaneous, intramuscular, and intradermal) carriers are interesting due to their many advantages, which include:

- Stabilization of antigen;
- Modulation of antigen release;
- Improved antigen city by adjuvant action.

### 5.2 Using microparticulate carriers for targeting:

Site-specific drug delivery, or targeting, is a well-established concept that is gaining a lot of interest. The efficacy of the medication as a treatment is determined by its capacity to selectively interact with and get access to its target receptors. The use of a carrier system mediates the drug action and enables the drug to leave the pool in a targeted, efficient, and repeatable manner

### 5.3 Targeting microspheres was made easier by monoclonal antibodies:

The microspheres that monoclonal antibodies target are known as immunological microspheres. This targeting allows for selective targeting to certain locations. Monoclonal antibodies are extremely selective molecules. Because of their excellent specificity, monoclonal antibodies (Mabs) can be utilized to target specific sites with microspheres containing bioactive substances. By Mab spheres and microspheres can be directly connected through covalent binding. The free aldehyde, amino, or hydroxyl groups on the surface of the microspheres can be bound by the antibodies. Maps can be added to microspheres using any of the methods listed below:

- Direct connection;
- Specific and nonspecific adsorption
- Reagent coupling.

## II. CONCLUSION:

Compared to many other forms of medication delivery systems, microspheres are a superior option. By combining a number of different approaches, microspheres will eventually play a key role in novel drug delivery, specifically in the sorting of diseased cells, diagnostics, gene and genetic materials, safe, targeted, specific, and efficient in vitro delivery, and supplements that function as tiny replicas of the body's diseased organs and tissues. Compared to current technology, microspheres have a number of advantages. These have become a fascinating new platform for biologists to use these methods when studying the relationships between biomolecules

and cellular functions. Microspheres have been employed in a growing number of research in recent years for a wider variety of purposes, and it is clear that the potential There are a ton of applications. These devices have been utilized in pulmonary perfusion and inhalation studies, as well as to scan the liver, brain, heart, and gastrointestinal tract. Although the name "microsphere" is brief, it has several uses in drug delivery systems to achieve desired biological activity. Microspheres will play a key role in innovative drug delivery systems by combining multiple approaches, namely in cell sorting, genetic engineering, and diagnostics. The study demonstrates that the innovative medication delivery system works well with microspheres as carriers.

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