

A Review on Self-Micro Emulsifying Drug Delivery System (SMEDDS)

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ABSTRACT: Self-Micro Emulsifying Drug Delivery System (SMEDDS) is a promising formulation approach for delivering poorly water-soluble drugs. SMEDDS are consist of oils, surfactants, co-surfactants, and water, which self-assemble into microemulsions upon contact with water. This review highlights the essential aspects of SMEDDS, including their preparation, characterization, advantages, limitations, and applications in the pharmaceutical industry. The significance of SMEDDS in enhancing the solubility and bioavailability of drugs has been discussed in detail, along with the prospects of this technology.

KEYWORDS: SMEDDS, bioavailability, solubility.

I. INTRODUCTION

Oral bioavailability of many drugs is often hindered by their poor solubility in water, which leads to low absorption and, consequently, limited therapeutic efficacy. This is a significant challenge in the development of effective drug formulations, especially for lipophilic (fat-soluble) compounds. When these drugs are administered orally, they may not dissolve adequately in the gastrointestinal tract,

I. SMEDDS: Composition and Working Mechanism^{4,5,6}

SMEDDS contains several components that work synergistically to enhance drug delivery. The basic ingredients of SMEDDS include:

- **Oils:** These are lipophilic substances that serve as the vehicle of drug. Common oils used in SMEDDS include medium-chain triglycerides (MCTs), long-chain triglycerides (LCTs), and natural oils.
- **Surfactants:** These compounds reduce the interfacial tension between the oil and water phases, aiding in the formation of microemulsions. Surfactants generally used in

which impedes their ability to reach the systemic circulation in sufficient concentrations to produce a desired effect¹.

To address these challenges, Self-Micro Emulsifying Drug Delivery Systems (SMEDDS) have been developed as a promising solution. SMEDDS are isotropic mixtures of oils, surfactants, and co-surfactants that, upon ingestion, form stable microemulsions in the gastrointestinal tract. This process helps enhance the solubility of poorly water-soluble drugs, facilitating their absorption. The system creates tiny droplets of oil in water, which significantly increases the surface area for drug absorption, improving bioavailability¹⁻².

These systems have gained significant attention in pharmaceutical research due to their ability to improve the dissolution and bioavailability of lipophilic drugs. SMEDDS offer a versatile approach for enhancing the therapeutic efficacy of drugs that would otherwise be poorly absorbed, making them a valuable tool in drug delivery systems. The development of SMEDDS has opened up new possibilities in formulating effective treatments for a variety of diseases, particularly those requiring drugs with low aqueous solubility³.

SMEDDS include polysorbates, polyethoxylated castor oil, and phospholipids.

- **Co-surfactants:** Co-surfactants assist in stabilizing the microemulsion system. Examples include alcohols like ethanol or propylene glycol.

When SMEDDS are introduced into the aqueous environment (such as the gastrointestinal tract), they spontaneously form a microemulsion due to the interaction of the surfactants with the aqueous phase. This process enhances the solubilization of poorly water-soluble drugs and improves their absorption in the GIT.

II. Preparation of SMEDDS⁶⁻⁹

The preparation of SMEDDS is a critical factor influencing the quality and stability of the final product. The preparation process typically involves the following steps:

1. **Selection of Components:** The first step in formulating SMEDDS is choosing the appropriate oil, surfactant, and co-surfactant based on the solubility properties of the drug.
2. **Screening of Surfactants and Co-Surfactants:** A series of experiments, such as phase diagrams, are performed to determine the ideal ratio of surfactant to co-surfactant which is likely to yield a stable microemulsion.
3. **Preparation of SMEDDS:** The oil, surfactant, and co-surfactant are mixed in specific ratios, followed by the addition of the drug. The mixture is typically prepared using methods like high-speed homogenization or solvent evaporation.
4. **Incorporation of Drug:** The drug is incorporated into the SMEDDS formulation by dissolving it in the oil phase or mixing during the preparation process.
5. When SMEDDS are introduced into the aqueous environment (such as the gastrointestinal tract), they spontaneously form a microemulsion due to the interaction of the surfactants with the aqueous phase. This process enhances the solubilization of poorly water-soluble drugs and improves their absorption in the GIT. The stability characteristics of the formulation are assessed through various tests such as temperature cycling, centrifugation, and visual inspection for phase separation.

Characterization of SMEDDS⁹⁻¹¹

The success of SMEDDS depends largely on their characterization, which ensures the quality and stability of the formulation. The common characterization techniques include:

- **Droplet Size and Distribution:** The droplet distribution in the microemulsion is an essential factor in ensuring that the drug can be effectively absorbed. Smaller droplets (less than 200 nm) provide a larger surface area for absorption, to improve bioavailability.
- **Zeta Potential:** Zeta potential indicates the stability of the microemulsion. A higher zeta potential (typically greater than ± 30 mV) is associated with stable emulsions.

- **Viscosity:** The viscosity of the formulation is important for ease of administration, especially for oral formulations.
- **In Vitro Drug Release:** The release profile of the drug from the SMEDDS is typically studied using dissolution studies, which simulate the release of the drug under gastrointestinal conditions.
- **Thermodynamic Stability:** To evaluate the long-term formulation stability, tests like freeze-thaw cycles, centrifugation, and temperature control cycles are performed.

Advantages of SMEDDS¹²⁻¹⁵

SMEDDS offer several advantages over traditional drug delivery systems, especially for poorly water-soluble drugs:

- **Improved Solubility:** The primary benefit of SMEDDS is the enhanced solubility of lipophilic drugs. The microemulsion formed during the digestion process significantly increases the surface area for drug absorption.
- **Increased Bioavailability:** By increasing solubility, SMEDDS can increase the bioavailability of drugs, especially those with poor aqueous solubility.
- **Ease of Preparation:** SMEDDS are easy to prepare and scale up, making them cost-effective for large-scale manufacturing.
- **Stability:** SMEDDS formulations are generally thermodynamically stable, which reduces the chances of drug degradation during storage.
- **Non-invasive Delivery:** SMEDDS can be formulated into oral, topical, and other non-invasive delivery systems, providing more options for drug delivery.

Limitations of SMEDDS¹⁴⁻¹⁷

Despite their many advantages, SMEDDS also have certain limitations:

- **Dependence on Food Intake:** Some SMEDDS formulations may yield improved outcomes when taken with food, as the lipid content may enhance the solubility of the drug.
- **Expensive Ingredients:** High-quality surfactants and co-surfactants used in SMEDDS can be expensive, which may increase the overall cost of the formulation.
- **Formulation Stability:** The stability of SMEDDS can be affected by factors such as temperature, storage conditions, and the presence of water, potentially leading to phase separation.

- **Complexity in Manufacturing:** While the formulation of SMEDDS is straightforward, scaling up the manufacturing process can be complex and requires precise control of formulation components.

7. Applications of SMEDDS¹⁴⁻¹⁷

SMEDDS are versatile systems that can be used in various therapeutic areas, including:

- **Cancer Therapy:** SMEDDS can be used to deliver anticancer agents, often poorly soluble in water, improving their therapeutic efficacy.
- **HIV Treatment:** SMEDDS have been used in the formulation of antiretroviral drugs, helping to increase their bioavailability.
- **Pain Management:** Drugs for pain relief, such as non-steroidal anti-inflammatory drugs (NSAIDs), can benefit from SMEDDS formulations by increasing their absorption and effectiveness.
- **Vitamins and Nutraceuticals:** SMEDDS are widely used in the formulation of vitamins and other nutritional supplements, increasing their bioavailability.

II. FUTURE PROSPECTS

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The development of SMEDDS is a scope of ongoing research. Recent advancements aim to improve the efficiency of SMEDDS by exploring new excipients, improving stability, and enhancing manufacturing processes. Furthermore, the application of SMEDDS for delivering biologics and peptides is gaining attention due to their potential to improve the oral bioavailability of these complex molecules.

In the future, the composition of SMEDDS with novel technologies such as nanoparticles, nanostructured lipid carriers (NLCs), and solid lipid nanoparticles (SLNs) could further enhance the performance of these systems.

III. CONCLUSION

Self-Micro Emulsifying Drug Delivery Systems (SMEDDS) have emerged as a beneficial approach in pharmaceutical formulations, offering significant advantages in delivering poorly water-soluble drugs. While challenges such as formulation stability and ingredient costs remain, SMEDDS continue to show promise in improving the solubility, bioavailability, and overall therapeutic effectiveness of drugs. Ongoing research and technological advancements will likely enhance their applications in the future.

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