

## “Development of Stability Indicating Assay Method for Estimation of Olmesartan Medoxomil and Chlorthalidone in Its Combined Dosage Form”

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### ABSTRACT:

A simple, rapid, accurate, precise and economical reverse phase high performance liquid chromatographic method was developed for estimation of two Anti-hypertensive drugs, Olmesartan medoxomil and Chlorthalidone. The separation of both the drugs was achieved on Phenomenex kinetex XB C-18 column (150 mm X 4.6 mm, 5 $\mu$  particle size) using mobile phase Acetonitrile : 0.1% TFA Water (45:55v/v). The flow rate was 1 ml/minute with run time 10 minutes and detection was done at 235 nm. The retention time of Olmesartan medoxomil and for Chlorthalidone was 2.15 min. and 1.71 min. respectively. The linearity of the method was evaluated at a range of 50 to 150  $\mu$ g/ml and 31.25 to 93.75  $\mu$ g/ml for Olmesartan medoxomil and Chlorthalidone respectively. The Correlation Coefficient of Olmesartan medoxomil and Chlorthalidone were 0.999 each. Precision studies were carried out and %RSD of peak areas of Olmesartan medoxomil and Chlorthalidone was about 0.051 and 0.077 respectively. The percentage recoveries of both the drugs OLM and CHL from the tablet formulation were 98.27% and 98.33% respectively. The developed method has been statistically validated according to ICH guidelines and found to be simple, cost-effective, linear, precise, and accurate with the prescribed values. Thus this method can successfully employed for simultaneous estimation of both drugs in commercial products.

**KEY WORDS:** Olmesartan medoxomil, Chlorthalidone, RP-HPLC, Stability.

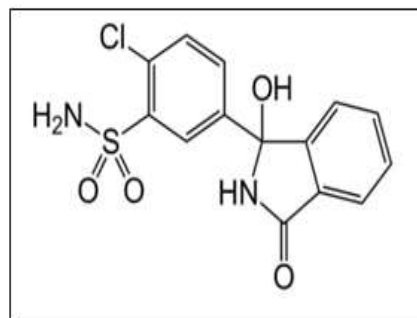
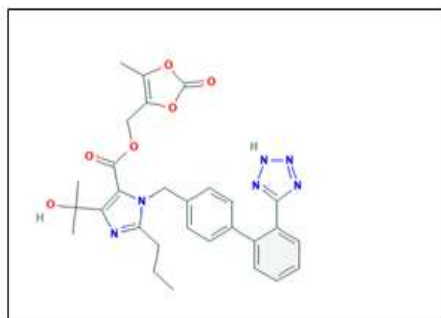
### I. INTRODUCTION:

Combinations of two or more drugs in the Pharmaceutical dosage forms are very much useful

in multiple therapies. Market survey revealed that, day-by-day new drugs and their combination with another drugs are being introduced in market as they have more patient compliance than a single drug<sup>1</sup>. The analytical chemistry hence has challenge in developing the methods for their analysis with the help of number of analytical techniques, which are available for the estimation of the drugs and their combination<sup>2</sup>. Analytical monitoring of pharmaceutical product or of specific ingredients within the product is necessary to ensure the safety and efficacy throughout the shelf life, including storage, distribution and use<sup>3</sup>. The Olmesartan medoxomil and Chlorthalidone fixed-dose combination is found to show superior antihypertensive efficacy in blood pressure reduction in patients with hypertension.

**Olmesartan Medoxomil** is an Angiotensin II receptor antagonist with the chemical name (5-methyl-2-oxo-2H-1,3-dioxol-4-yl)methyl-4-(2-hydroxypropan-2-yl)-2-propyl-1-({4-[2-(2H-1,2,3,4-tetrazol-5-yl)phenyl]phenyl)methyl}-1H-imidazole-5-carboxylate. It is BCS class II drug (low solubility, high permeability) & having the white to light yellowish-white crystalline appearance.<sup>18,19</sup>

**Chlorthalidone** is a diuretic agent involved in the treatment of hypertension. It is a sulfonamide derivative with different chemical structure from thiazide but the same pharmacological actions as that of thiazide diuretic with the chemical name 2-chloro-5-(1-hydroxy-3-oxo-2,3-dihydro-1H-isindol-1-yl) benzene-1-sulfonamide. It is BCS class IV drug (low solubility, low permeability) which is having white or yellowish-white crystalline appearance.<sup>18,19</sup>



**Figure no. 1:** Structure of Olmesartan medoxomil **Figure no. 2:** Structure of Chlorthalidone

The present work was undertaken to develop such method of analysis, which can estimate both the drugs in combination which is a precise, accurate, simple, reliable and less time consuming method for estimation of drugs in tablet.

## II. EXPERIMENTAL:

### MATERIALS & METHODS:

#### Material and Reagents:

Olmesartan medoxomil and Chlorthalidone was provided by Aarti Pharmaceuticals, (Mumbai, India). Pharmaceutical formulation of tablets containing Olmesartan medoxomil 20 mg and Chlorthalidone 12.5 mg was purchased from local commercial sources. HPLC grade acetonitrile, water and methanol was procured from Fisher scientific chemicals, Mumbai and TFA was procured from Loba chemie, Mumbai. All solutions were filtered through a Millipore vacuum filter system (0.22 µm) and degassed by the sonicator.

#### Instrument:

A HPLC system consisted of LC solution data handling system (HPLC SYS-LC-138 Systronics), 80 plus intelligent UV/VIS detector, micro-syringe SGE analytical science; Australia was used for injecting the samples. Data acquisition were performed by using Clarify software. Ultrasonic bath sonicator was used for

degassing of the mobile phase. Weighing of the materials was carried out on a Mettler balance with high degree of accuracy.

#### Methods:

#### Determination of wavelength for detection of OLM and CHL

The suitable wavelength for detection of OLM and CHL was selected from overlain spectrum of OLM and CHL, wavelength selected was 235 nm.

#### Selection of mobile phase

Based on sample solubility, stability, suitability various mobile phases and compositions were tried to get a good resolution and sharp peak.

The following mobile phases were tried:

1. Methanol : Water [40 : 60 %v/v]
2. Acetonitrile : Water [30 : 70 %v/v]
3. ACN : Methanol : Water [30 : 10 : 60 % v/v]
4. ACN : 0.1% TFA Water [35 : 65 %v/v]
5. ACN : 0.1 % TFA Water [45 : 55 %v/v]

Each mobile phase was filtered through Millipore vacuum filter system (0.22 µm) and degassed by sonication for 20 minutes. From the various mobile phases tried, mobile phase containing ACN and 0.1 % TFA Water in 45: 55 %v/v proportion was selected, since it gave completely resolved peaks with symmetry within limits and significant retention times for both the drugs.

#### Chromatographic conditions:

**Table no.1:** Chromatographic conditions.

Chromatographic mode	Chromatographic conditions
Stationary phase	Phenomenex Kinetex XB-C18 (150mm x 4.6 mm, 5µ)
Standard solution	100 µg/ml OLM & 100 µg/ml CHL
Column Oven Temp:	30 <sup>0</sup> C

Mobile Phase	ACN : (0.1%) TFA Water (45:55 v/v)
Flow rate	1 ml/min.
Sample size	10µl
Run Time	10 minutes
Detection Wavelength	235nm
Diluent	ACN : Water (50:50)

**Selection of chromatographic conditions**

The selection of RP-HPLC method for separation depends upon the nature of the sample, it's molecular weight and solubility. 27 The chromatographic variables such as mobile phase, flow rate and solvent ratio were studied. The resulting chromatograms were recorded and the chromatographic parameters such as selectivity, asymmetric factor, and peak resolution were calculated. The condition that gave the best resolution, symmetry and selectivity was selected for estimation.

**Optimization of Chromatographic Parameters**

Optimisation in HPLC is the process of finding a set of conditions that adequately analyze the quantification of the analyte with acceptable accuracy, precision, sensitivity, specificity, cost, ease, and speed of analysis.

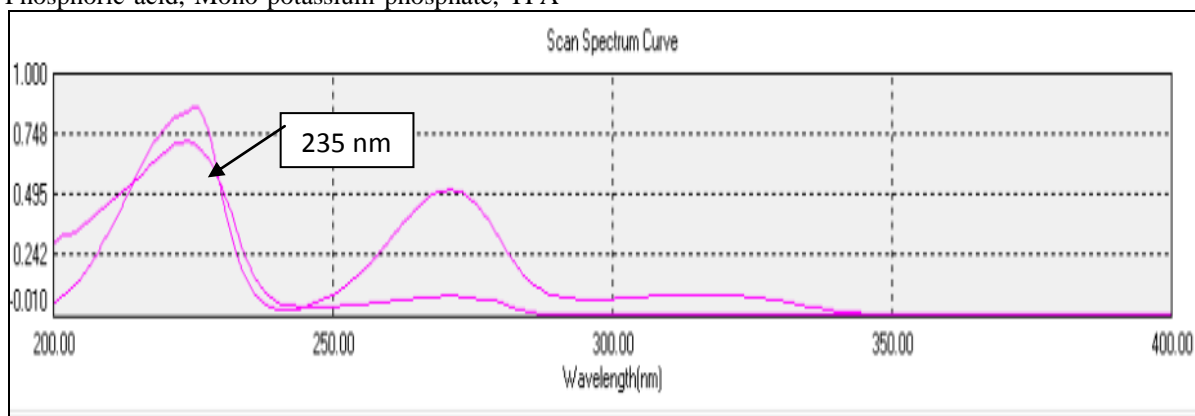
**Optimization of Mobile phase strength:**

Different trials and error are carried out on Water, Methanol and Acetonitrile by changing different mobile phase ratio and different stabilizers to resolve both peaks. Stabilizers like Ortho Phosphoric acid, Mono potassium phosphate, TFA

etc. Mobile phase is optimized as the minimum quantity of buffer and Acetonitrile should be used to carry out separation of peak of both the drugs because buffer may damage the column chemistry as they bind most to bonded phase and Acetonitrile is best solvent for column but that is too expensive, so to develop precise an economical method their concentrations are minimized. TFA (Trifluoroacetic acid) was used in mobile phase to make low p<sup>H</sup> buffer. TFA is more advantageous over all other acid as it improves resolution and sharpens the peak; it is volatile, easily removed and it is having low absorption within detection wavelength. Finally, mobile phase containing Acetonitrile and 0.1 % TFA Water in 45:55 % v/v proportions was found to give best resolution for both the drugs.

**Selection of wavelength maxima:**

Olmesartan Medoxomil showed absorption maxima at 255 nm and Chlorthalidone showed at 268 nm. For simultaneous estimation a common wavelength i.e. isobestic point for detection was selected at 235 nm which is shown in **Figure no. 3.**



**Figure no. 3:** UV Spectra to determine Isobestic point.

**Preparation of mobile phase:**

450 ml of Acetonitrile & 550 ml of 0.1% TFA HPLC grade water measured with help of measuring cylinder and transferred to 1000 ml of volumetric flask. Then this mobile phase is sonicated for 10 minutes.

**Standard and working standard:**

Accurately weighed 10 mg of standard Olmesartan medoxomil and Chlorthalidone and dissolved into 100 ml mobile phase to obtain the final concentration of 100 µg/ml. Appropriate dilutions were made into 10 ml volumetric flask with the mobile phase to produce working solutions in the concentrations range 50-150 µg/ml and 31.25-93.75 µg/ml for Olmesartan medoxomil and Chlorthalidone respectively. 10µl of samples were injected into the chromatographic system and peak areas were measured.

**Sample solution:**

The sample solution of formulation product is prepared by taking average of the 20 tablet and crushed into a mortar by using the pestle and accurately weighing of equivalent weight of powdered tablet containing 10 mg of OLM and 6.25 mg of CHL and dissolved in 10ml of diluent, shake it for 5 minutes for complete dissolution of the powdered material and then 0.45 micron filter was used to remove insoluble material from

solution and then sonicated for 5 minutes and then 1 ml of above solution was further diluted to 10 ml of diluent, Finally the remaining volume is make upto the mark by mobile phase. Injected 10 µl of filtered portion of the sample and standard preparation into the chromatographic system. Record the responses for the major peaks. Calculated the content of Olmesartan medoxomil and Chlorthalidone present in each tablet.

**Linearity Study of Drug at Selected Wavelength:**

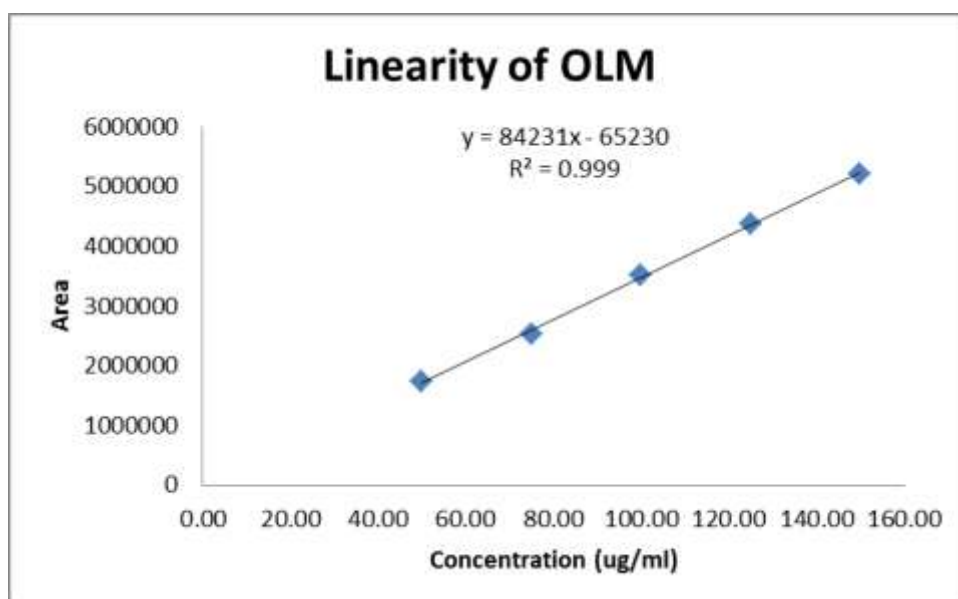
Appropriate aliquots of STD Stock Solutions of OLM and CHL were prepared in combined form and transferred to series of 10ml volumetric flasks respectively. Volume made to the mark to get Solution of drug in different concentrations level at 50-150 µg/ml of OLM & 31.25-93.75µg/ml of CHL. 10µL injection of each concentration of the drug were injected into HPLC system three times separately and chromatogram was obtained. Evaluation of the drug was performed with the UV detector set as 235 nm and peak area Vs concentration response was recorded. According to ICH Guidelines n>5 i.e. using minimum 5 concentrations (50,75,100,125,150) were taken. A linear relationship evaluated across the range of analytical procedure. Which is shown in **Table no. 2 & 3.**

**Table no. 2:** Linearity of OLM at 235 nm.

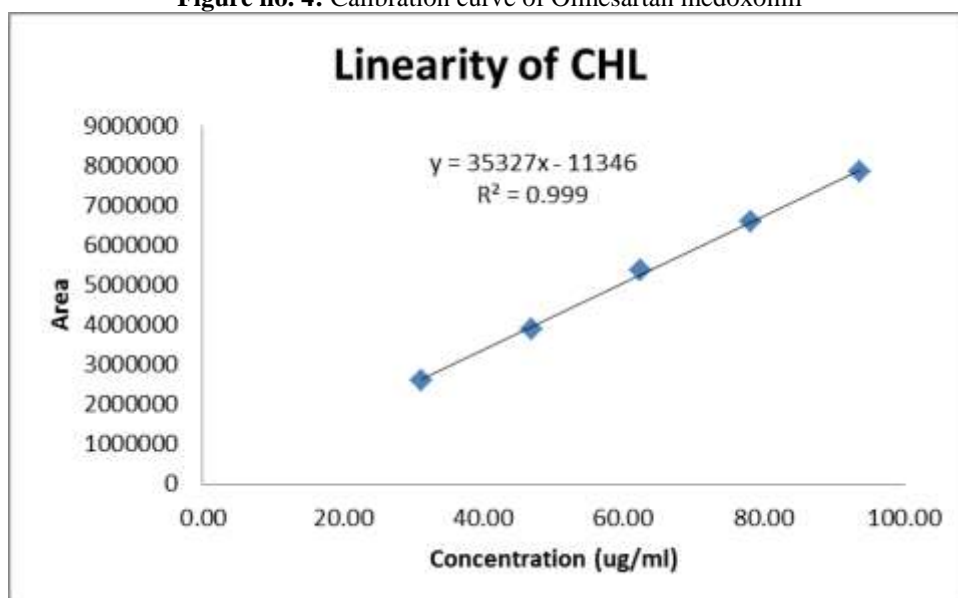
Sr.no.	Conc. (µg/ml)	Peak 1	Peak 2	Peak 3	Average	SD	%RSD
1	50	2601	2603	2604	2602	1.527	0.058706
2	75	3885	3887	3885	3886	1.154	0.029714
3	100	5360	5363	5361	5361	1.527	0.028493
4	125	6577	6574	6577	6576	1.732	0.026339
5	150	7838	7839	7838	7838	0.577	0.007366
<b>Avg.</b>						1.303	0.0301

**Table no.3:** Linearity study of Chlorthalidone at 235 nm

Sr. no.	Conc. (µg/ml)	Peak 1	Peak 2	Peak 3	Average	SD	%RSD
1	31.25	1723	1724	1725	1724	1.00	0.058005
2	46.88	2515	2518	2518	2517	1.732	0.068814
3	62.50	3515	3512	3514	3513	1.527	0.043482
4	78.13	4367	4366	4364	4366	1.527	0.034987
5	93.75	5218	5217	5215	5216	1.527	0.029285
Avg.						1.462	0.04691



**Figure no. 4:** Calibration curve of Olmesartan medoxomil



**Figure no. 5:** Calibration curve of Chlorthalidone

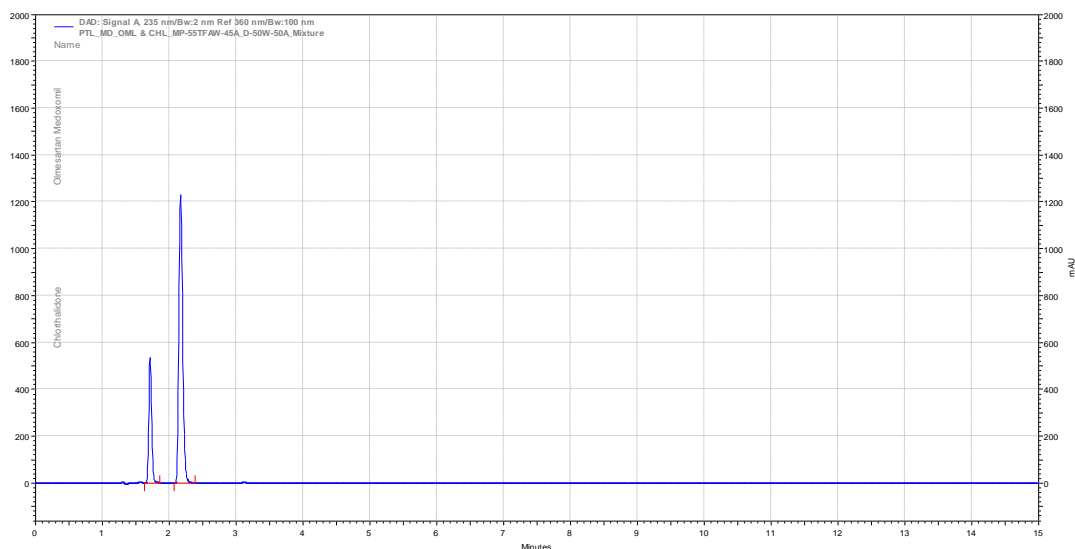


Figure no. 6: Chromatogram of OLM & CHL

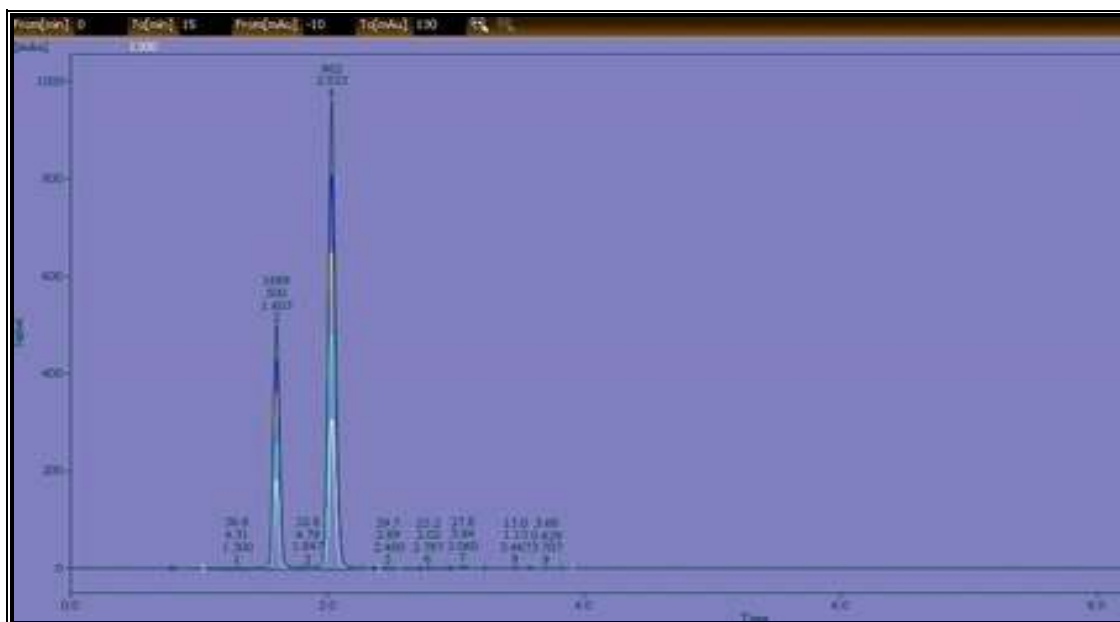


Figure no. 7: Overlain graph of OLM and CHL

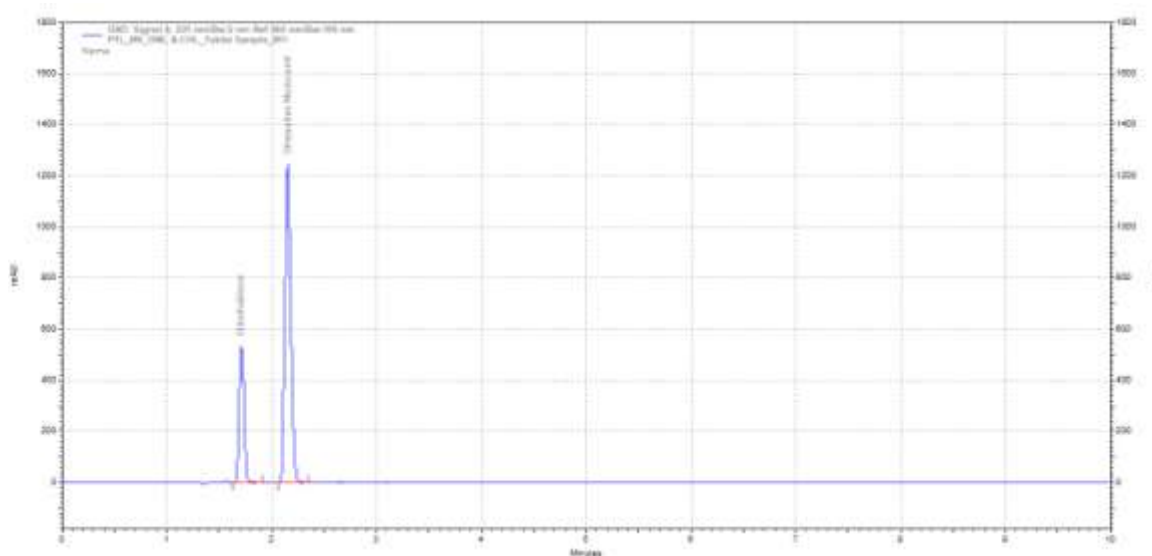
**Analysis to Marketed Formulations:-**

For the present work Olmesar CH-20 (Macleods) has 20 mg OLM and 12.5 mg CHL was selected for the analysis. The percentage drug content was calculated by measuring the peak area

and comparing with the pure OLM and CHL of respective concentration. The statistical data obtained after replicate determinations (n = 6) is shown in **Table no. 4.** and chromatogram is shown in **Figure no. 8.**

**Table no. 4:** Marketed formulation analysis

Formulation Olmesar CH-20 (Macleods)	Label claim (mg/tab)	Conc. µg/ ml	Area	Amount found µg/ml	Amount found (mg/tab)	%Assay= [sample area/std area]*100
<b>OLM</b>	20	998	10183158	100.11	20.06	100.11
<b>CHL</b>	12.5	625	3477630	99.86	12.41	99.86



**Figure no. 8:** Typical Chromatogram of marketed formulation

**Recovery Studies:**

Accuracy of analysis was determined by performing recovery studies by spiking different concentrations of pure drug in the preanalyzed

tablet sample. Results of recovery studies indicated that the method is rapid, accurate and reproducible and is shown in **Table no 5**.

**Table no. 5:** Data of Recovery study of OLM & CHL

	OLM				CHL			
Level	Conc.	% REC	Mean of % recovery	%RSD	Conc.	% Recovery	Mean of % recovery	%RSD



75%	20	96.68	98.27	0.63	12.5	95.54	98.33	0.51
100%	20	99.99		0.11	12.5	100.02		0.33
125%	20	98.16	98.28	0.62	12.5	99.43		0.52

\*Obtained from 6 determinations, S.D: Standard deviation, R.S.D: Relative standard deviation.

### System Suitability Parameters

System suitability parameters were analyzed on freshly prepared standard stock solutions of OLM and CHL. Both the drugs were injected into the chromatographic system under the optimized chromatographic conditions. Parameters were validated according to ICH Q2B guidelines for validation of analytical procedures in order to determine the linearity, sensitivity, precision and accuracy for the analyte. The suitability of the system were

- a) Retention time.
- b) Number of theoretical plates.
- c) Tailing/ asymmetric factor.
- d) Limit of Detection
- e) Limit of quantitation
- f) Resolution
- g) Calibration Range ( $\mu\text{g ml}^{-1}$ )

The results are shown in **Table no. 6.**

**Table no 6:** System Suitability Parameters

Sr. No.	Parameters	OLM	CHL
1	Retention time	2.15min	1.71min
2	Theoretical plate	6587.4	6732
3	Asymmetry	1.25	1.078
4	LOD	9.42	2.67
5	LOQ	28.54	8.08
6	Resolution	4.61	0.0
7	Calibration Range ( $\mu\text{g ml}^{-1}$ )	50-150	31.25-93.75

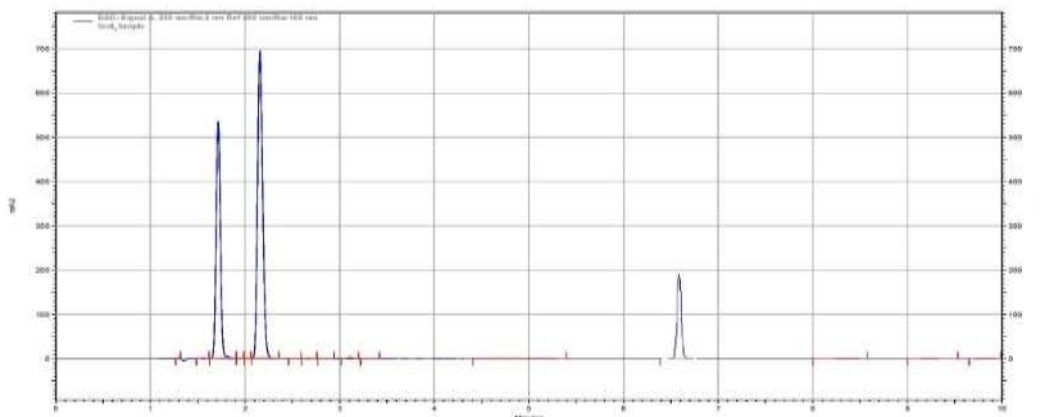
### Stability Study:

#### Acid Hydrolysis:

For acid hydrolysis 10 ml of 0.1N HCL in 10 ml sample solution containing 20 $\mu\text{g/ml}$  and 12.5 $\mu\text{g/ml}$  of Olmesartan medoxomil and Chlorthalidone was mixed separately and was kept at 60°C in water bath for 8hr. Then it diluted to 10

ml. Further 1ml was pipette out from both stock solutions and was diluted with diluent (50: 50, Water: ACN), Filter and injected into system and chromatogram were recorded to assess the stability of sample. The chromatogram was represented as **Figure no. 9.**



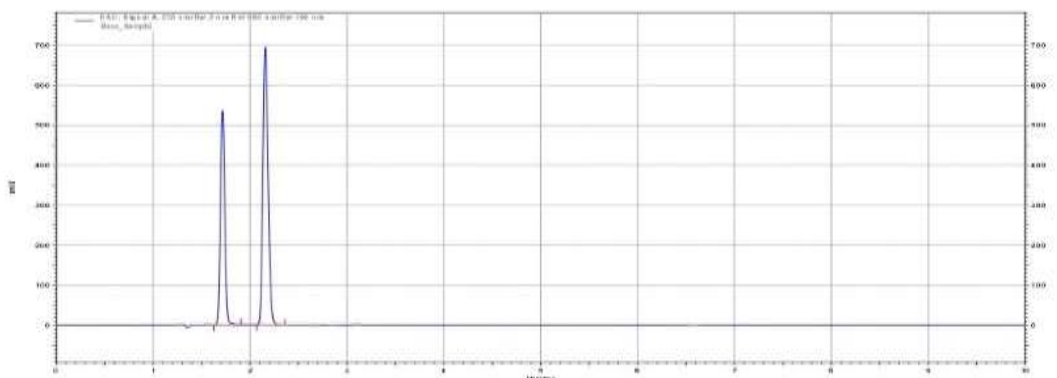


**Figure no 9:** Chromatogram of 0.1N Acid degradation of OLM and CHL

**Alkali hydrolysis:**

For alkali hydrolysis, 10 ml of 0.1 N NaOH was mixed to both the flask containing 10 ml sample solution of 20 $\mu$ g/ml and 12.5 $\mu$ g/ml of Olmesartan medoxomil and Chlorthalidone resp. and was kept at 60°C in water bath for 8hr. Further

1 ml was taken from both stock solutions and was diluted with diluent (50: 50, Water: ACN), Filter and injected into system and chromatogram were recorded to assess the stability of sample. The chromatogram were represented as **Figure no. 10**.



**Figure no. 10:** Chromatogram of 0.1N NAOH degradation of OLM and CHL

**Oxidative:**

To 10 ml of 20 $\mu$ g/ml Olmesartan medoxomil and 12.5 $\mu$ g/ml Chlorthalidone containing sample solution was taken separately in 10 ml volumetric flask. 10 ml of 3% v/v H<sub>2</sub>O<sub>2</sub> solution added to both the flask and was kept at 60°C in water bath for 10 min. Further mixed to 10

ml. Further 1 ml was taken from both stock solutions and was diluted with diluent (50: 50, Water: ACN), Filter and injected into system and chromatogram were recorded to assess the stability of sample. The chromatogram were represented as **Figure no. 11**.

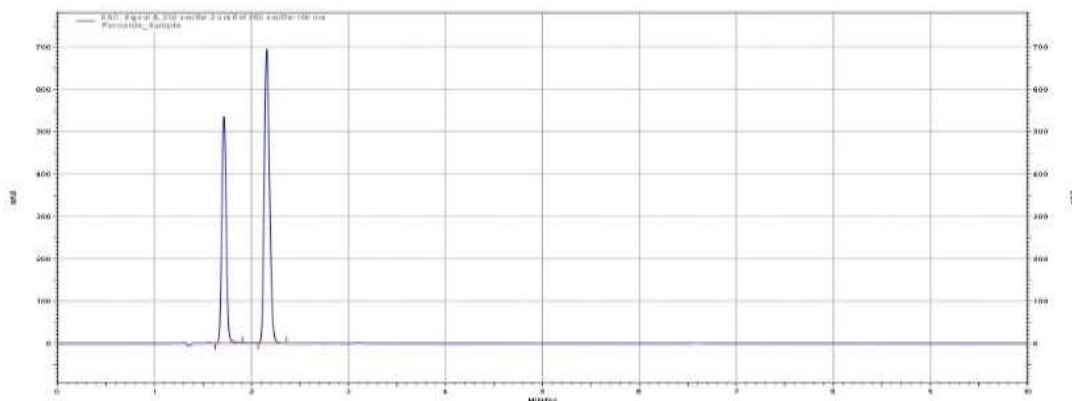


Figure no. 11: Chromatogram of 3% H<sub>2</sub>O<sub>2</sub> degradation of OLM and CHL.

**UV Analysis:**

10 ml of 20µg/ml Olmesartan medoxomil and 12.5µg/ml Chlorthalidone containing sample solution was taken separately in 10 ml VF. Then it was placed in UV cabinet at 254 nm for 5 hours and later was diluted with diluent. Further 1 ml was

taken from both stock solutions and was diluted with diluent (50: 50, Water: ACN), Filter and injected into system and chromatogram were recorded to assess the stability of sample. The chromatogram is represented as **Figure no. 12**.

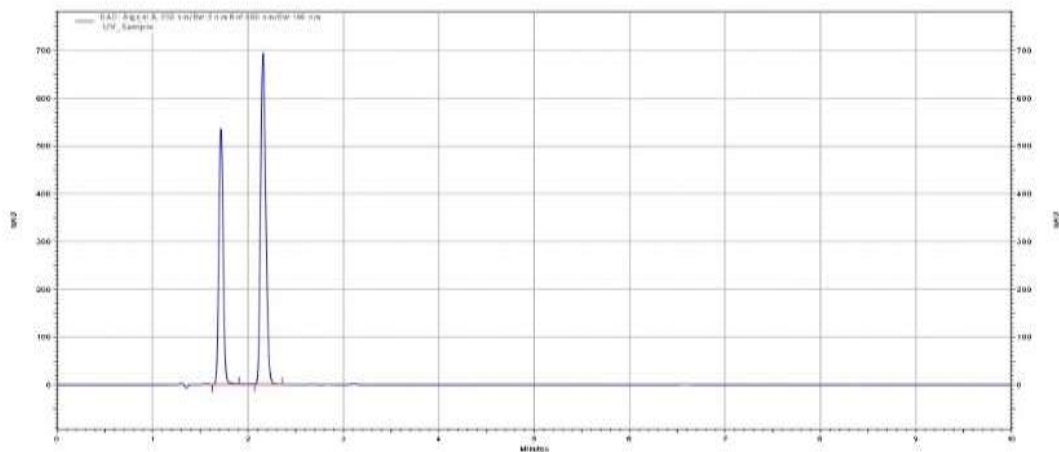


Figure no. 12: Chromatogram of Photolytic degradation of OLM and CHL

**Dry Heat:**

10 ml of 20µg/ml Olmesartan medoxomil and 12.5µg/ml Chlorthalidone containing sample solution was taken separately in 10 ml volumetric flask. Then it was kept in Oven at 60°C for 2 hours and later was diluted with diluent. Further 1 ml was

taken from both stock solutions and was diluted with diluent (50: 50, Water: ACN), Filter and injected into system and chromatogram were recorded to assess the stability of sample. The chromatogram is represented as **Figure no. 13**.

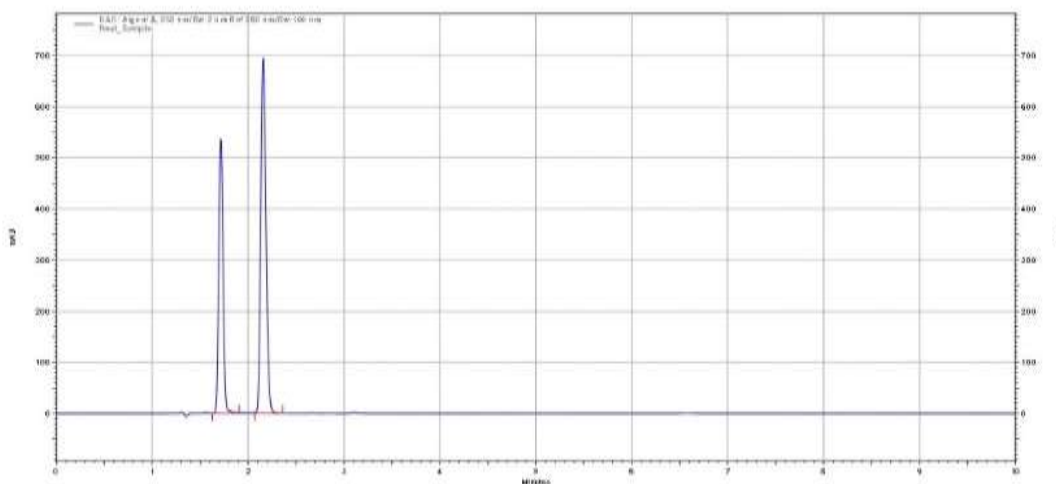


Figure no. 13: Chromatogram of thermal degradation of OLM and CHL

Table no.7: Results of % Degradation of OLM & CHL

Parameters	OLM			CHL		
	% assay	% degradation	% RSD	% assay	% degradation	% RSD
Acid	82.44	17.56	0.391	73.85	26.15	0.582
Base	98.70	1.30	0.514	94.36	5.64	0.638
Oxidative	95.13	4.87	0.652	97.01	2.99	0.531
Thermal	95.42	4.58	0.611	91.26	8.74	0.125
Photolytic	96.53	3.47	0.696	98.30	1.70	0.512

### III. RESULTS & DISCUSSION:

#### Development of Validated RP-HPLC Method for the Simultaneous Estimation of OLM and CHL in Combine Tablet Dosage Form

The separation was achieved on Phenomenex Kinetex XB-C18 (150mm x 4.6 mm, 5µ) Column using (0.1% TFA) water and Acetonitrile (55:45, v/v) as mobile phase for assay. Flow rate 1ml/min with sample size 10 µL & run time 10 minutes. Detection was carried out using UV detector at 235.0 nm. The retention time is 2.15 min. for OLM & 1.71 min. for CHL. The system suitability test shows the response with retention time, theoretical plate, tailing factor and peak area for both the drugs.

The proposed method when was applied for tablet dosage form the % label claim of OLM and CHL was found to be in limit respectively. The linearity obtained for OLM and CHL is in the range

between 31.25 to 93.75 mg/mL to 50 to 150mg/mL respectively. Accuracy of the method was checked by recovery studies at three different levels i.e. 75%, 100%, 125%. The % RSD values were less than 2 indicative of accuracy of the method. The method was found to be precise as indicated by the inter-day, intra-day analysis showing the % RSD values less than 2. Data from system suitability studies indicates conformity to compendia requirements. The method was also found to be robust as indicated by % RSD values less than 2 according to ICH guidelines.

#### Development of stability Profile of OLM and CHL in Combine Tablet Dosage Form.

For developing the stability profile the drug product is exposed to degradation medium like Acid, Alkali, Peroxide, Thermal & Photolytic. The parent drugs and degradation products formed

were well resolved under optimized chromatographic conditions. The % of degradation for OLM was 17.56 % (Acid), 1.30 % (Base), 4.87% (Peroxide), 4.58% (Thermal), 3.47 % (Photolytic) While for CHL 26.15 % (Acid), 5.64 % (Base), 2.99% (Peroxide), 8.74 % (Thermal), 1.70 % (Photolytic) which is in agreement with ICH Limit. It concludes that OLM and CHL both are more labile to acid.

#### IV. CONCLUSION:

The developed HPLC method was found to be more accurate, precise, economic, fast, reproducible and selective for simultaneous determination of OLM and CHL in combined tablet formulation. The statistical parameter of this method showed good results. The recovery studies revealed excellent accuracy and high precision of the method. The HPLC method was found to give better results. Therefore the proposed method could be applied for routine analysis in quality control laboratories. The stability study also gave very prominent results which could be used for routine analysis.

#### V. ACKNOWLEDGEMENT:

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#### REFERENCES:

- [1]. Skoog D.A., Holler F.J., Crouch S.R. Principle of Instrumental analysis. 6<sup>th</sup> edition , Thomsan publications, India ; 2007:300
- [2]. Skoog D.A, and Leary J.J. Principles of Instrumental Analysis, 4<sup>th</sup> ed. Philadelphia; Saunders College of Publishing; 1992: 1-2.
- [3]. Willard HH, Merritt LL, Dean JA, and Settle FA. Instrumental Methods of Analysis, 7<sup>th</sup> ed. U.S.A; Wadsworth Publishing Company; 1986: 5-97.
- [4]. Khopkar S.M. Basic concept of analytical chemistry, 2<sup>nd</sup> edition, New age International Ltd. Publisher, New delhi ;1998.
- [5]. Mohaammad Yunoos et.al Optimization and validation of RP-HPLC stability indicating method for simultaneous determination of Olmesartan medoxomil and chlorthalidone in pure drug and pharmaceutical dosage form , Journal of Chemical and Pharmaceutical Research,2015 ;7(3) : 2440-2448.
- [6]. Sharma BK. "Instrumental Methods of Chemical Analysis" 18th ed. Goel Publishing House, Meerut. 1997: page no. 3-10, 17, 70-71, 81-87.
- [7]. Azim Md. Sabir et al. "HPLC method development and validation: A Review" International Research Journal of Pharmacy;2013; 4 (4); 39-46.
- [8]. E. Kats. Quantitative Analysis Using Chromatographic Techniques. Wiley India pvt. Ltd.2009:193-211.
- [9]. Chatwal, GR, Anand, SK. Instrumental Methods of Chemical Analysis. 5th ed. Himalaya Publishing House, Mumbai 2002 : 2.149-2.171.
- [10]. K. A. Connors, Liquid Chromatography, A Textbook of Pharmaceutical Analysis 3<sup>rd</sup> edition, Willey Inter science, New York 1999 : 373-438.
- [11]. Beckett A.H., Stenlate J.B. Practical Pharmaceutical Chemistry, 4<sup>th</sup> edition CBS Publishers and distributors, New Delhi Part-2; 2002.
- [12]. Ahuja S. and M.W. Dong. Handbook of Pharmaceutical Analysis by HPLC, 1<sup>st</sup> edition ;2005:56.
- [13]. Sethi P.D. HPLC- Quantitative Analysis of Pharmaceutical Formulations , 1<sup>st</sup> edition, CBS publishers and distributors, New Delhi 2001 : 1-19.
- [14]. C.D. Gary. Analytical chemistry 5<sup>th</sup> edition, John Willey and Sons, Inc. 2001:1-3.
- [15]. P. W. Scott. Liquid Chromatography Coloumn Theory John Willey and Sons, Chichester 2001, Page No. 1-13.
- [16]. ICH Guidelines , Q1 A R2 Stability Testing Of New Drug Substance and Products International Conference on Harmonization, IFPMA, Geneva, 6 Feb,2003.
- [17]. Santosh k. Bhardwaj, et.al. A Review: HPLC Method Development and Validation, International Journal of Analytical and Bioanalytical Chemistry, 2015;5(4):76-81
- [18]. Rajani Pathuri, et.al. A review on analytical method development and

- validation of pharmaceutical technology , Andhrapradesh-521230
- [19]. Drug profile for Olmesartan Medoxomil is available from : <http://www.drugbank.ca/drugs/DBSALTO01815>
- [20]. Drug profile for Chlorthalidone is available from : <http://www.drugbank.ca/drugs/DB00310> Drug profile for OLM is available from: <https://pubchem.ncbi.nlm.nih.gov/130881> Drug profile for Chlorthalidone is available from : <https://pubchem.ncbi.nlm.nih.gov/2732>
- [21]. Sharma B.K. Instrumental method of chemical analysis, 23<sup>rd</sup> edition ,Goal publishing House, Meerut 2004.
- [22]. Cristian G.D. Analytical Chemistry, 5<sup>th</sup> edition, John Wiley and Sons, Inc., 2003.
- [23]. ICH guidelines. ICH, Q2 (RI) validation of analytical procedures: Text and methodology , Geneva Nov. 2005
- [24]. M.S. Aryane, et.al. HPLC Method development review, Journal of Chromatographic Science, 2010;Vol.48 : 721-725.
- [25]. Rajendra Patil, Tushar Deshmukh, et.al. Review on Analytical Method Development and Validation, 2014; 3(3).
- [26]. Phani.R.H , K.R.S. Prasad and UseniI Reddy Mallu, High resolution RP-HPLC method for the determination of hypertensive drug products, International Journal of Pharma and Bio Sciences ; 2013: 440 – 454.
- [27]. Manoj Charde, A. S. Welankiwar and Jitendra Kumar. Method development by Liquid Chromatography with Validation ,International Journal of Pharmaceutical Chemistry 2014; 4 (2): 57-61
- [28]. Pasbola, K., & Chaudhary.M. Updated Review on analytical method development & validation by HPLC. World Journal of Pharmacy & Pharmaceutical Sciences. 2017;6(5):1612-1630
- [29]. P. Murali et al. Development and validation of a stability-indicating RP-HPLC method for the simultaneous quantification of Olmesartan Medoxomil and Chlorthalidone in solid dosage form, Der Pharmacia Lettre, 2014; 6(6):148-157.
- [30]. Vilas Sawale et al. Development and Validation of RP-HPLC Method for the Simultaneous Estimation of Olmesartan Medoxomil and Chlorthalidone in Tablet dosage form,International Journal of Pharmacy and Pharmaceutical Sciences, 2015; 7(5): 266-269
- [31]. Jitendra Kumar et al. Development of forced degradation studies of drugs International Journal of Advances in Pharmaceutics 2013; 2(3): 34-40.

**List of abbreviations used:**

- 1 **CHL:** Chlorthalidone
- 2 **OLM:** Olmesartan Medoxomil
- 3 **Abs:** Absorbance
- 4 **Conc.:** Concentration
- 5 **S.D.:** Standard Deviation
- 6 **R.S.D.:** Relative Standard Deviation.
- 7 **H.P.L.C.:** High Performance Liquid Chromatography.